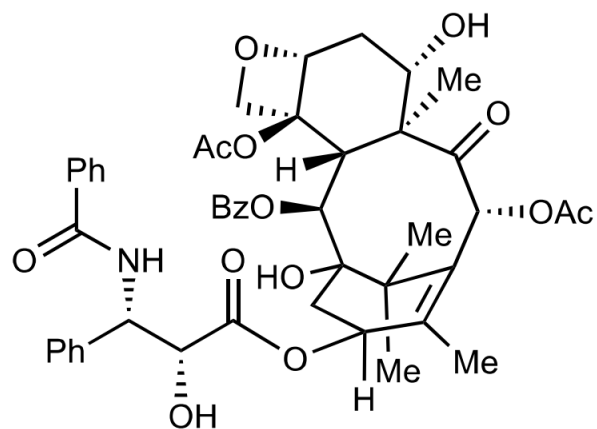


Taxane

MaTianhao

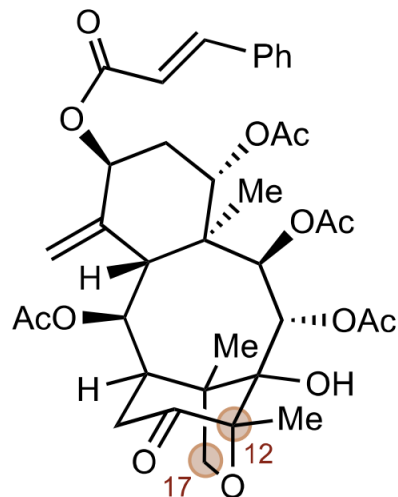
2022.11.24

The separation and biological activity



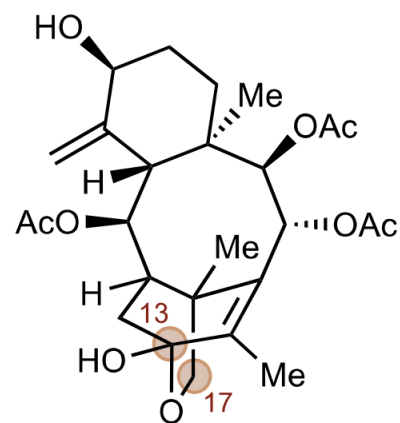
taxol (1)

FDA-approved anticancer agent



taxagifine (2)

C17-C12 oxo-bridged



taxezopidine A (3)

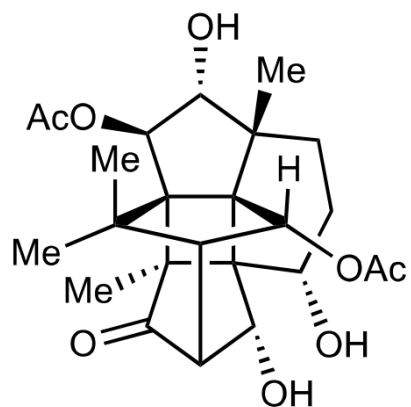
C17-C13 oxo-bridged

- In 1966, chemical investigation resulted in a highly active agent, isolated by **Wall** and **Wani** in a yield of **0.014%** and named it as taxol (1).
- Between the 1960s and 1970s, Japanese and American chemists isolated more than 20 taxanes from Japanese and European yews and most of them are **derivatives** of taxinine.
- 1974, Full clinical potential of taxol was **recognized** when the antitumor models changed to B16 melanoma against rapidly growing tumors and L1210 leukemia against solid tumors in 1974.

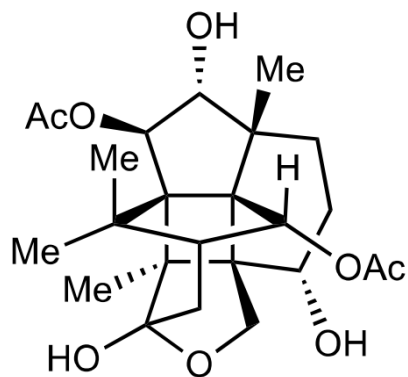


The Pacific yew *T. brevifolia*

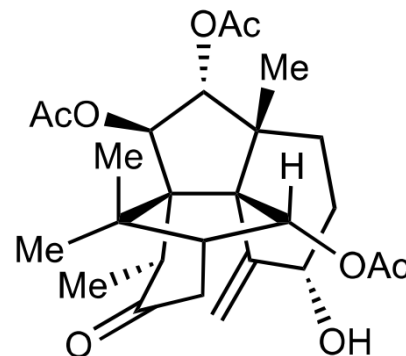
The separation and biological activity



canataxpropellane (4)



taxpropellane (5)



taxinine K (6)

•In 2007, A novel taxane canataxpropellane (4) with an unprecedented 5/5/4/6/6/6-membered hexacyclic skeleton containing [3.3.2]propellane was isolated from the needles of the Canadian yew, *Taxus canadensis*. by Kiyota group.

•It suffers from extremely inefficient sourcing from its natural producer, thus preventing biological and pharmaceutical investigations to this day



Canadian yew, *Taxus canadensis*



Richmond Sarpong

Macalester College, St. Paul, MN.

Bachelor of Arts Degree (Chemistry Major), May 1995 (Advisor Prof. Rebecca C. Hoye)

Princeton University, Princeton, NJ.

Master of Science Degree (Organic Chemistry), May 1997

Ph. D. in Organic Chemistry, May 2001 (Advisor Prof. Martin F. Semmelhack)

California Institute of Technology, Pasadena, CA

UNCF•Pfizer Postdoctoral Fellow, 2001–2004 (Advisor Prof. Brian M. Stoltz)

Professional Experience

Assistant Professor, Department of Chemistry (July 2004–2010) UCB

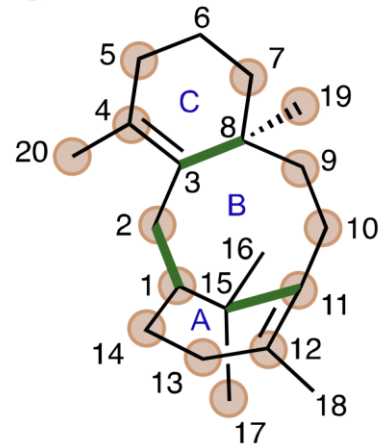
Associate Professor, Department of Chemistry (July 2010–June 2014) UCB

Full Professor, Department of Chemistry (July 2014–Present) UCB

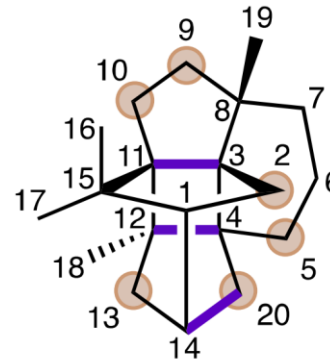
Retrosynthetic Analysis

● potential oxidations

● transannulations

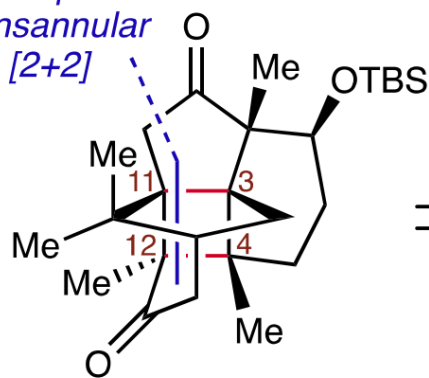


classical taxane (8)
6/8/6 ring system

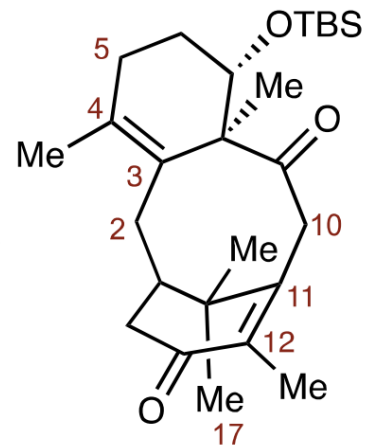


nonclassical taxane (9)
additional transannular
C-C bonds

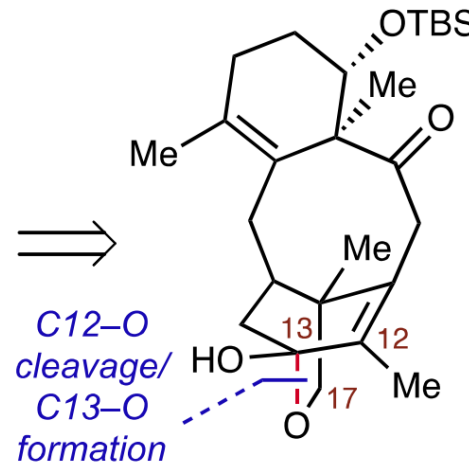
bioinspired
transannular
[2+2]



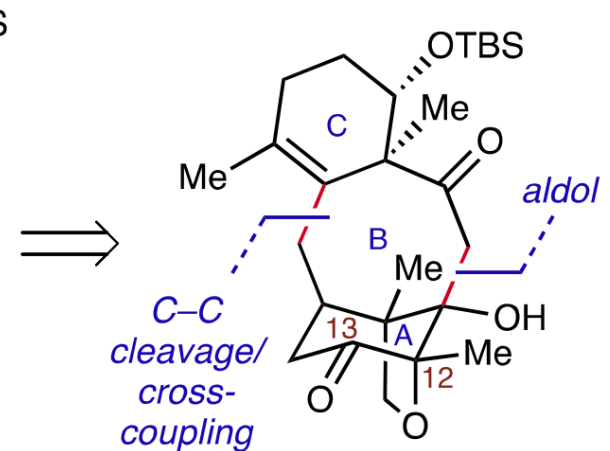
cyclotaxane core (13)



taxol core (14)

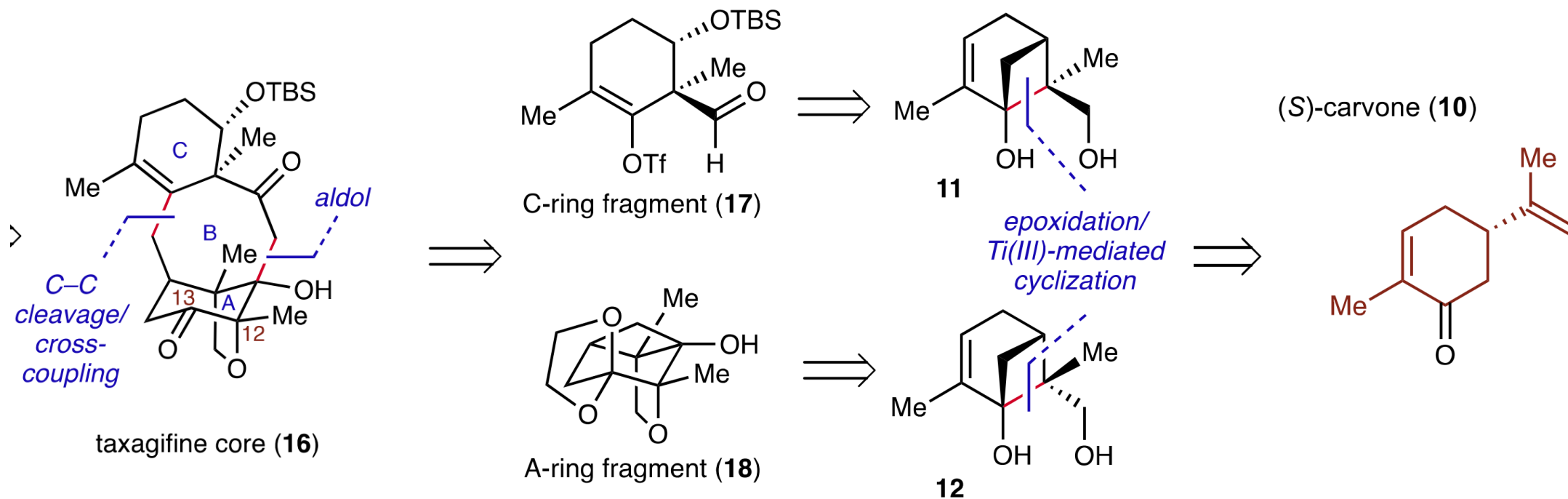


taxezopidine A core (15)

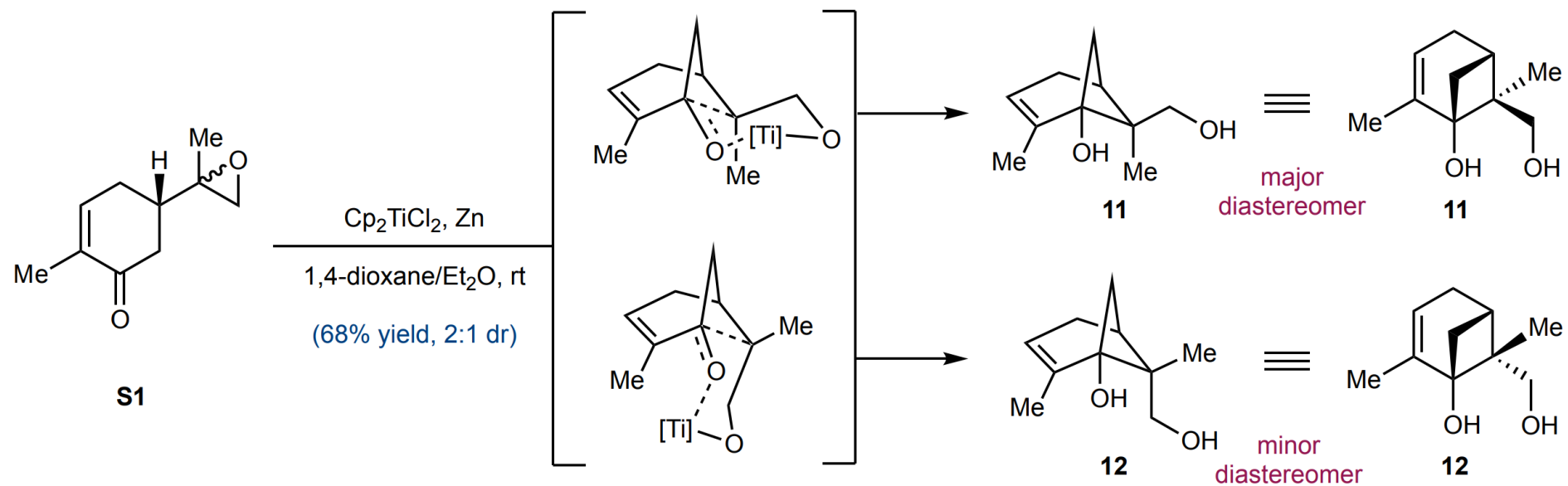
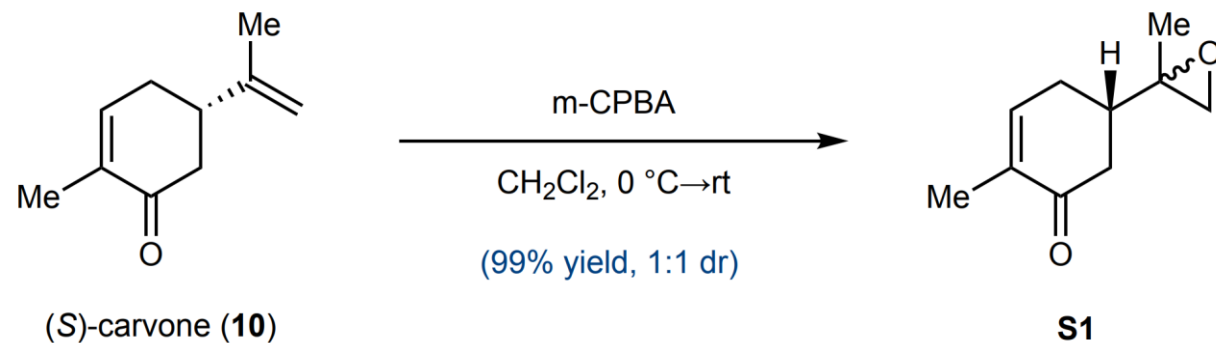


taxagifine core (16)

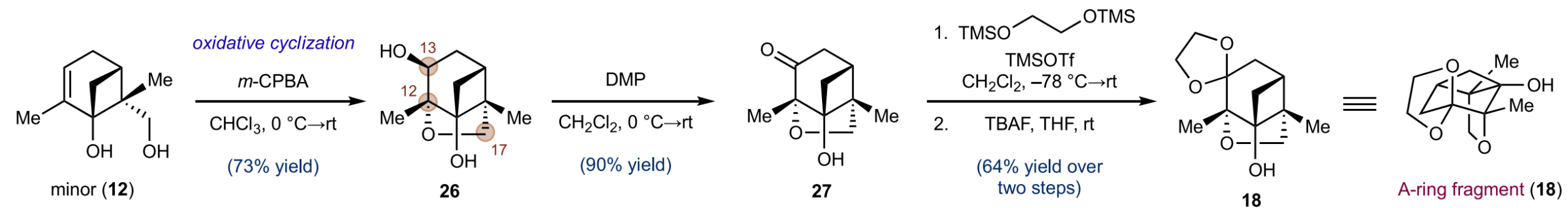
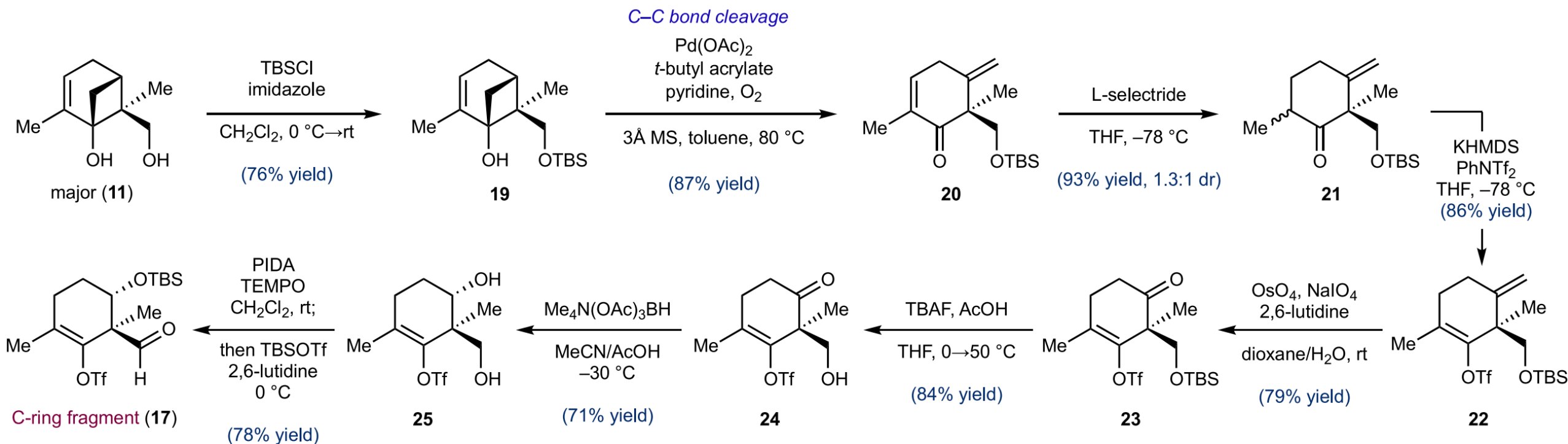
Retrosynthetic Analysis



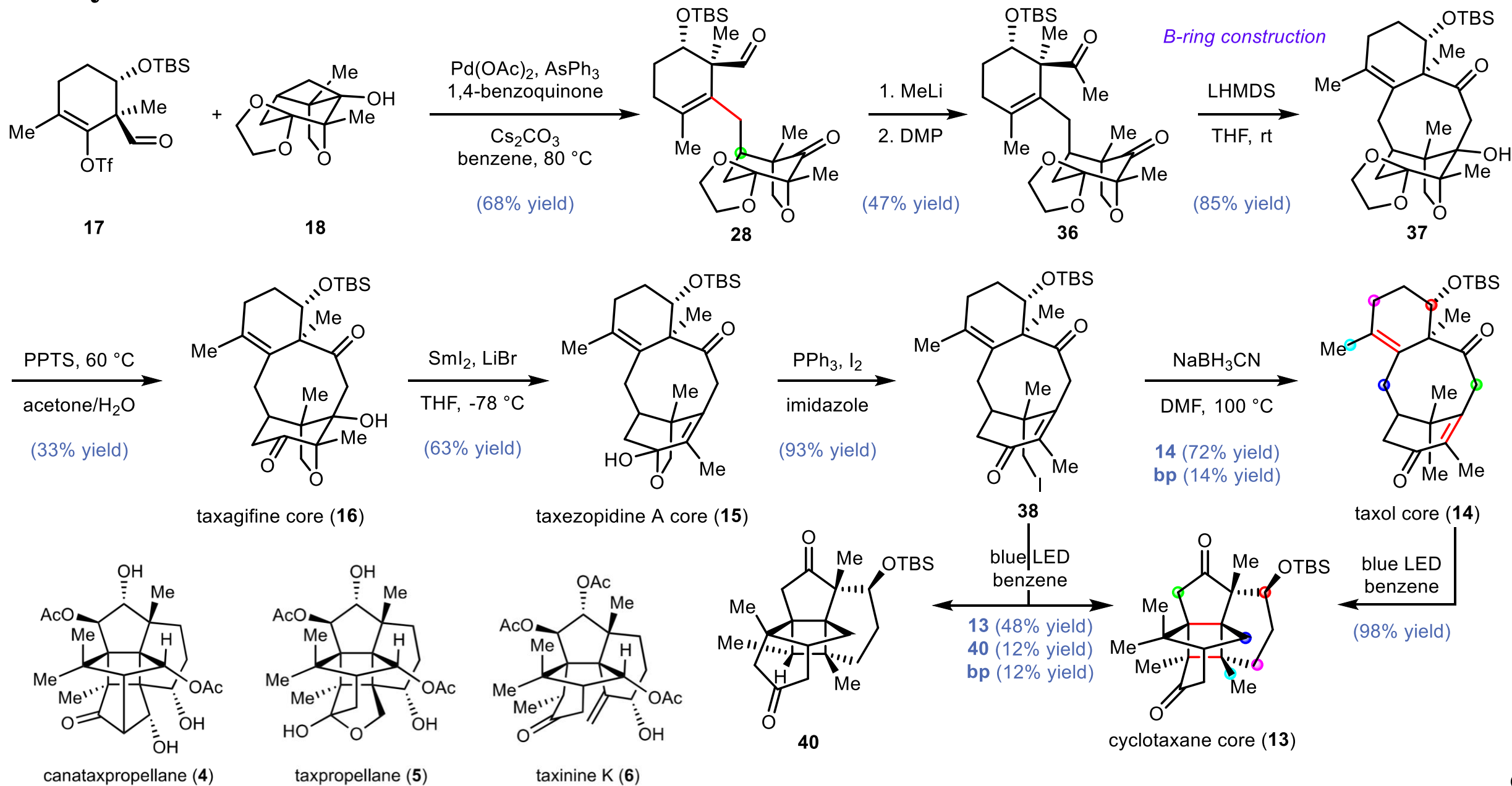
Synthesis



Synthesis



Synthesis





Tanja Gaich

Studies of Molecular Biology (University of Salzburg), 1998-2000

Studies of Chemistry (University of Vienna), 1999-2005

PhD with Prof. J. Mulzer (University of Vienna), 2005-2009

FWF-Postdoctoral fellow with Prof. P. S. Baran (Scripps. La Jolla CA, USA), 2009-2010

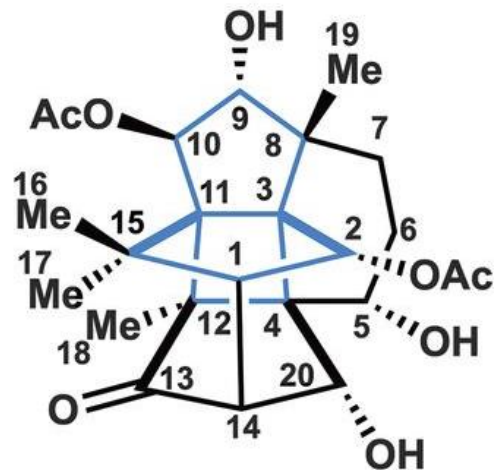
Professional Experience

Independent Researcher under supervision of Prof. M. Kalesse (Leibniz University Hannover), 2010-2015

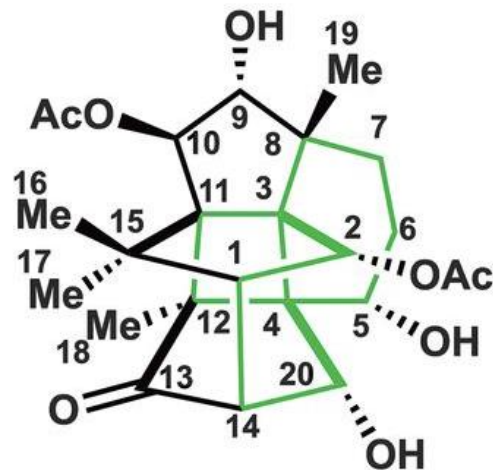
Full Professor of Organic Chemistry (University of Konstanz), 2015-present

Structural features

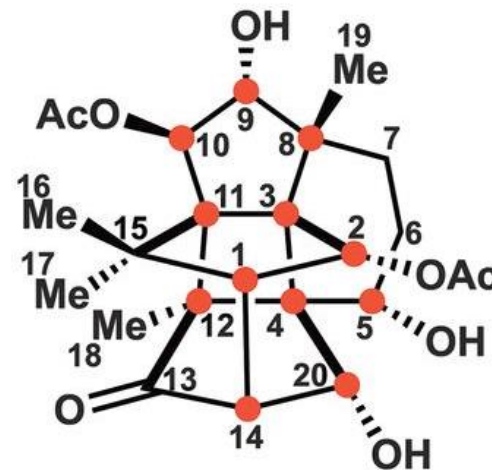
C Structure analysis



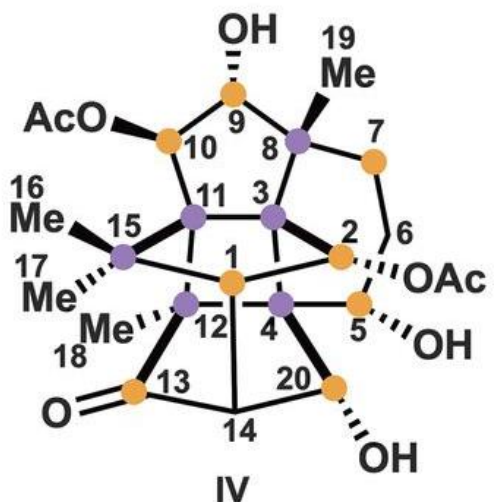
I



II



III

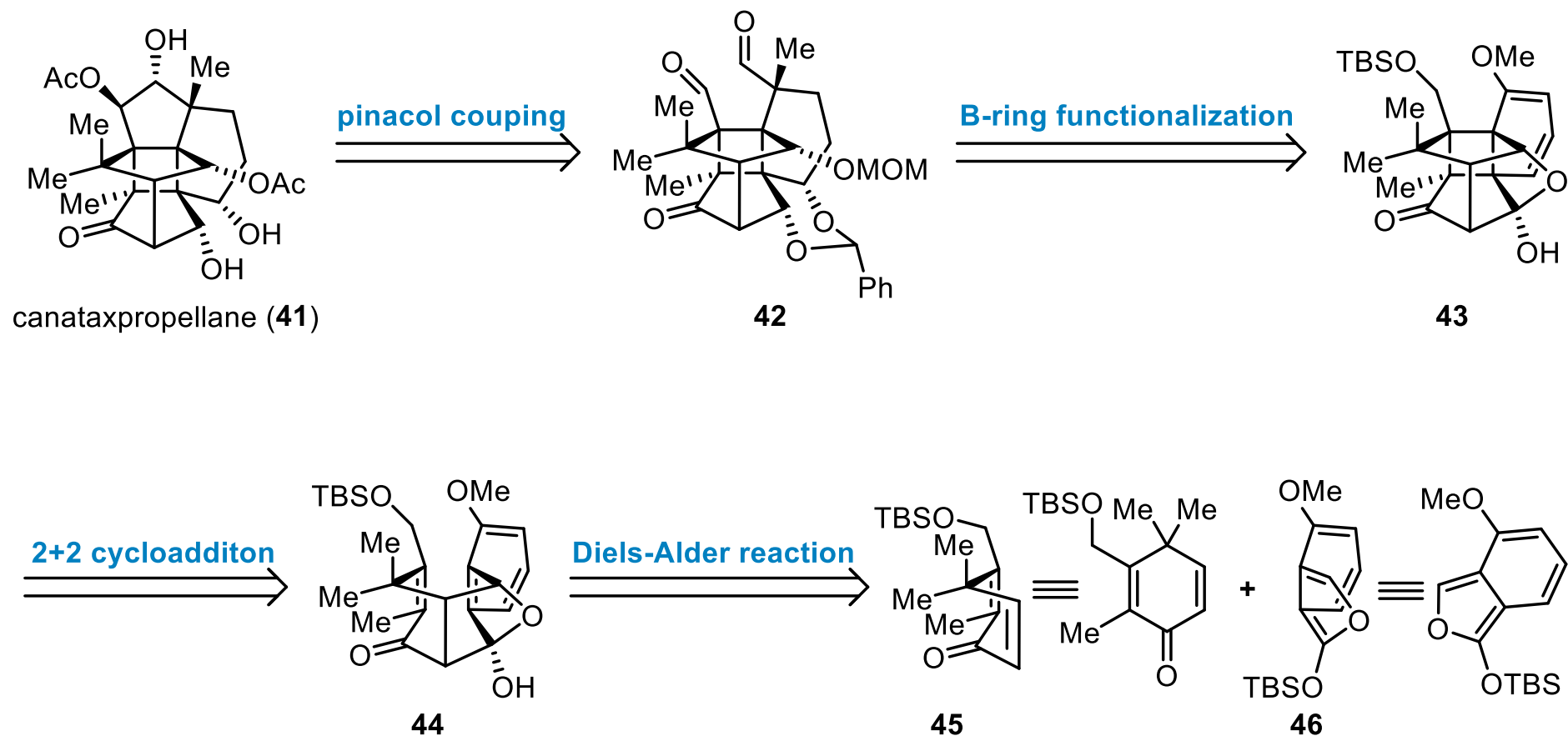


IV

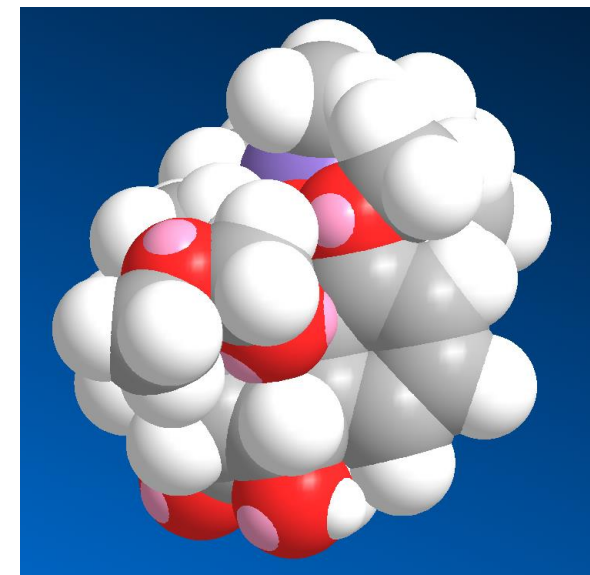
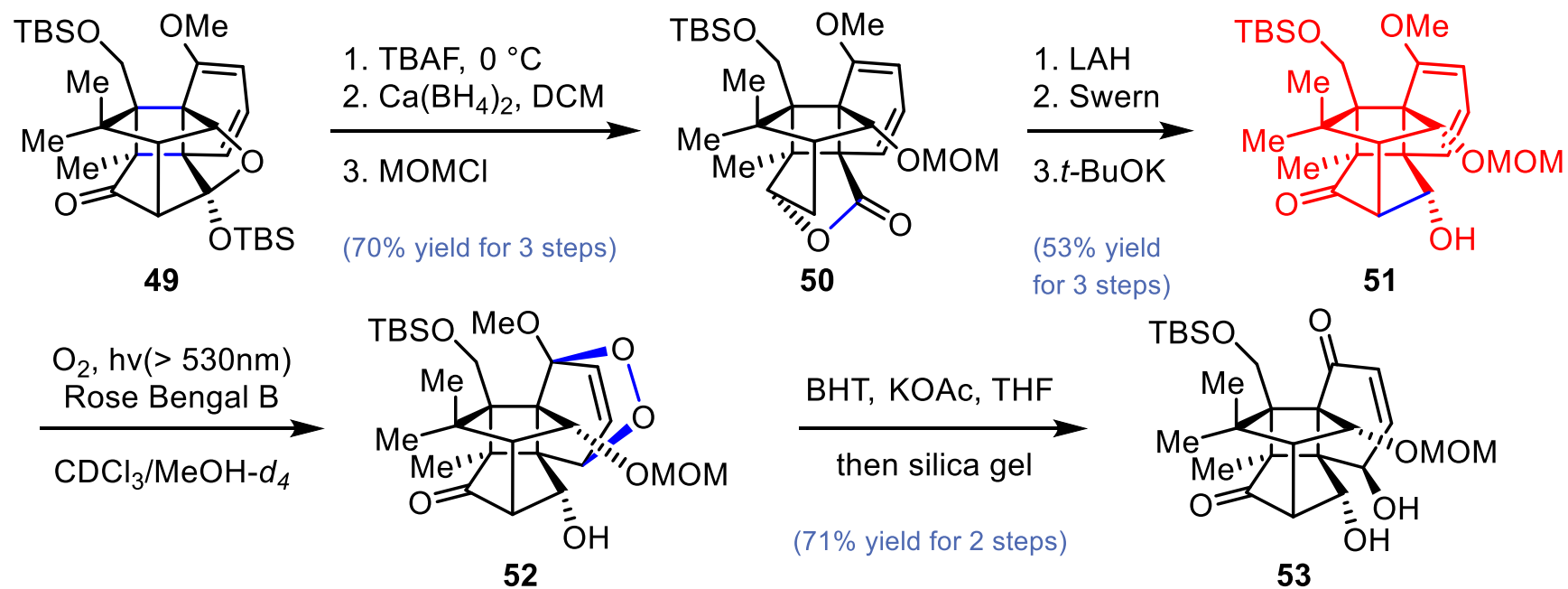
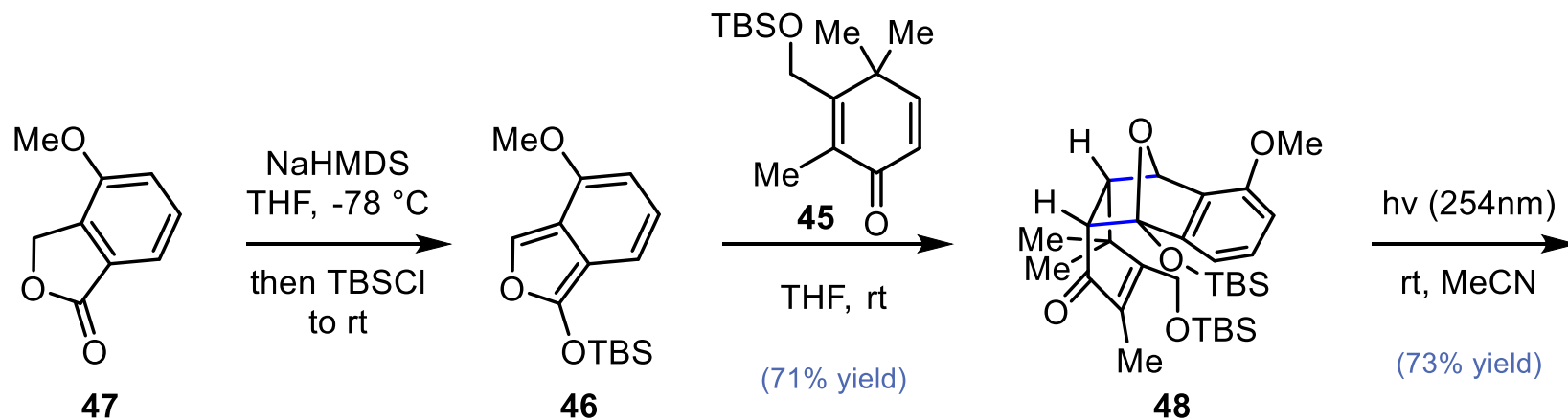
Structural features

- [3.3.2]propellane (blue I)
 - [4.4.2]propellane (green II)
 - 12 contiguous stereocenters (red III)
 - cyclobutane with all quaternary stereocenters (purple IV)
 - 6 quaternary carbons (contiguous; purple IV)
- 5 quaternary stereocenters (contiguous 3; 4; 8; 11; 12 IV)
 - 8 neopentyl positions (yellow IV)
 - only 2 non-neopentyl positions (6; 14 IV)
 - densely functionalized & highly oxidized

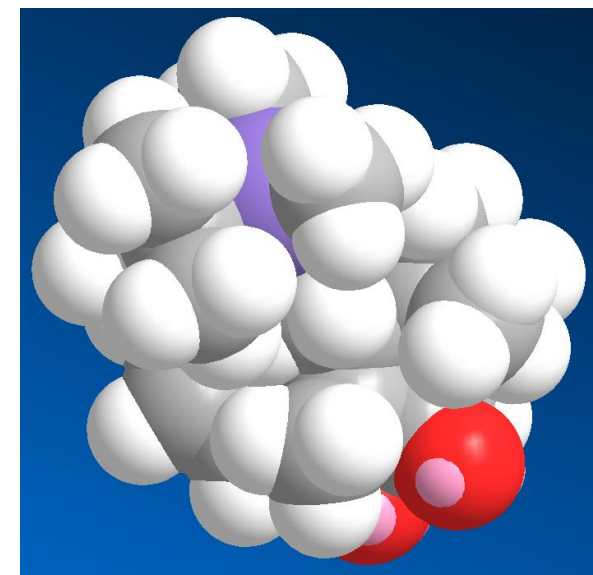
Retrosynthetic Analysis



Synthesis

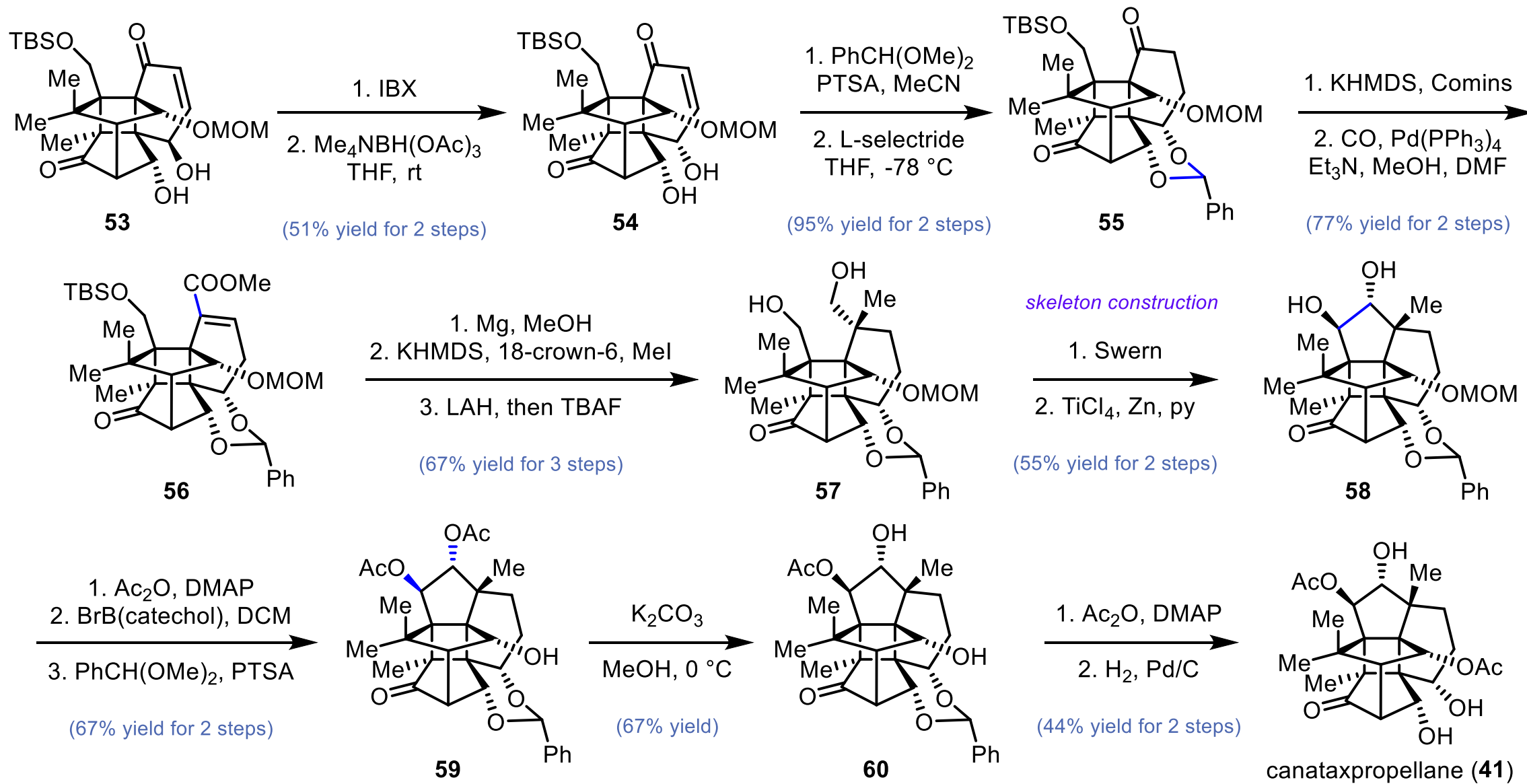


51 exo face (see from right)

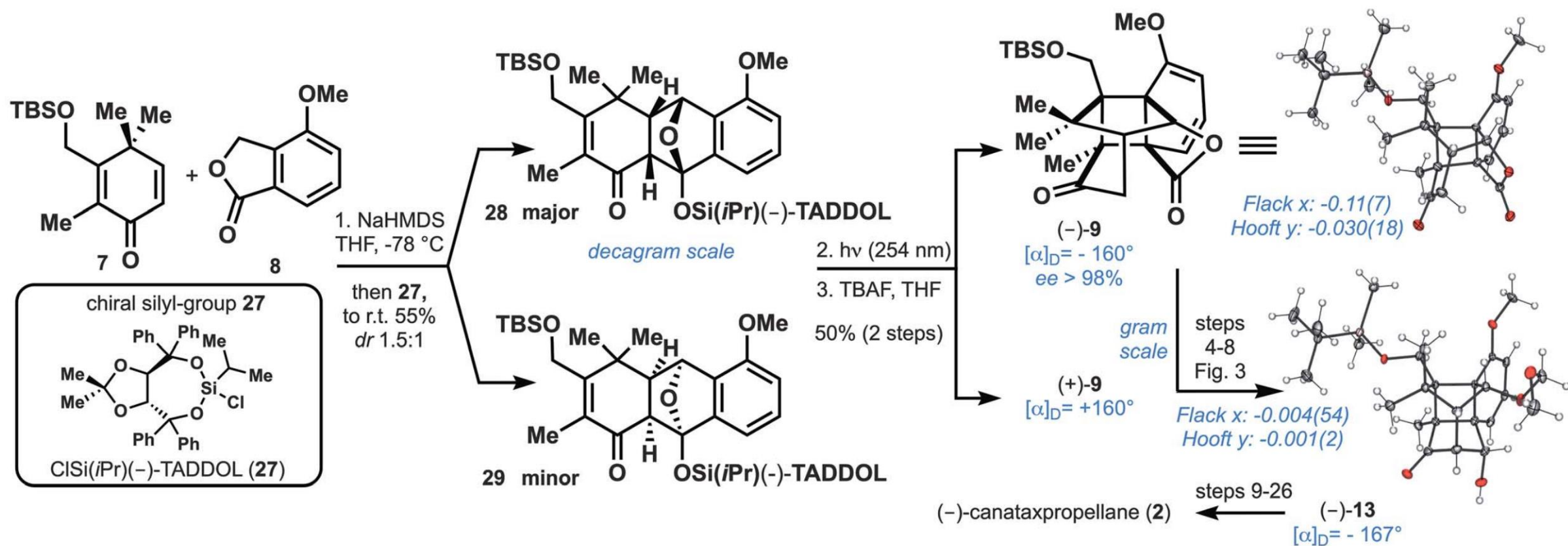


51 endo face (see from left) 13

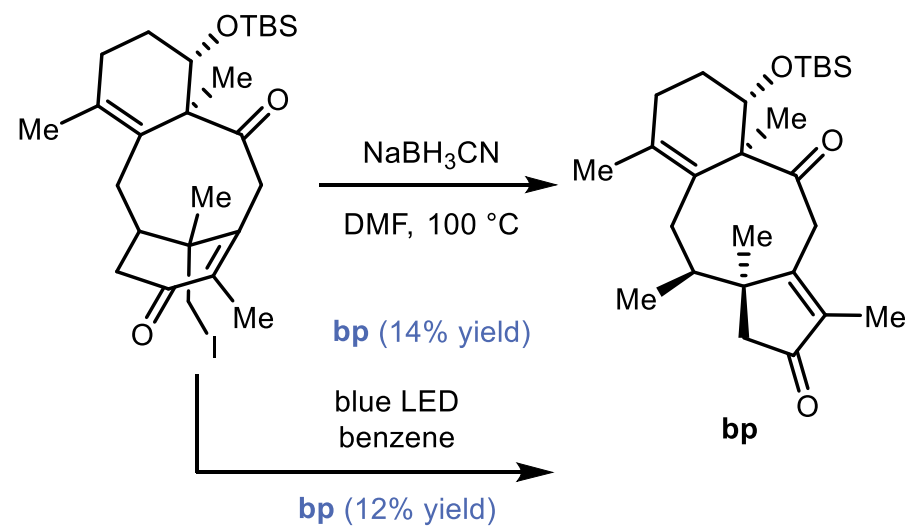
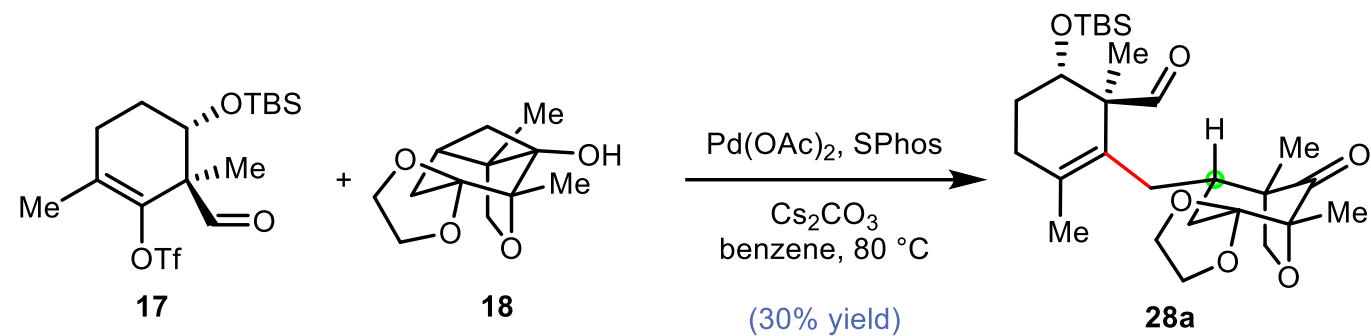
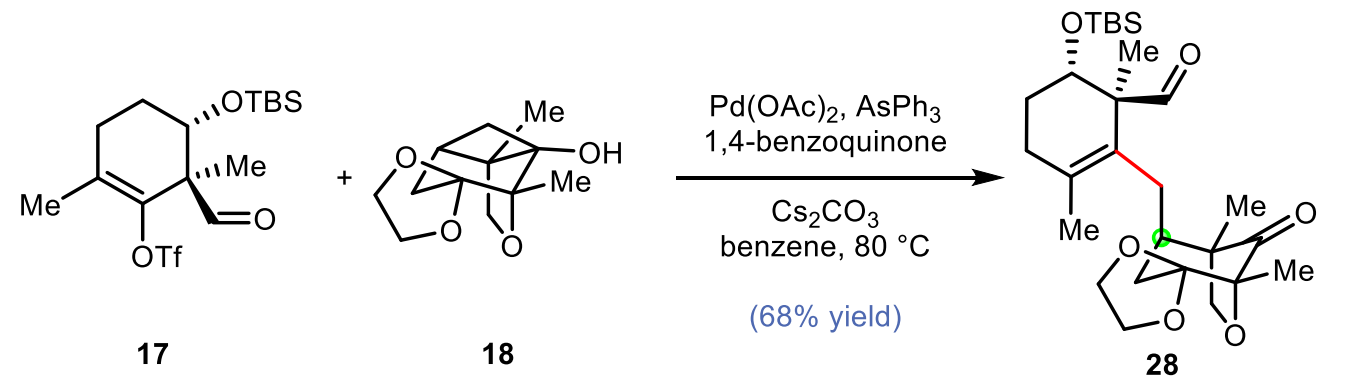
Synthesis

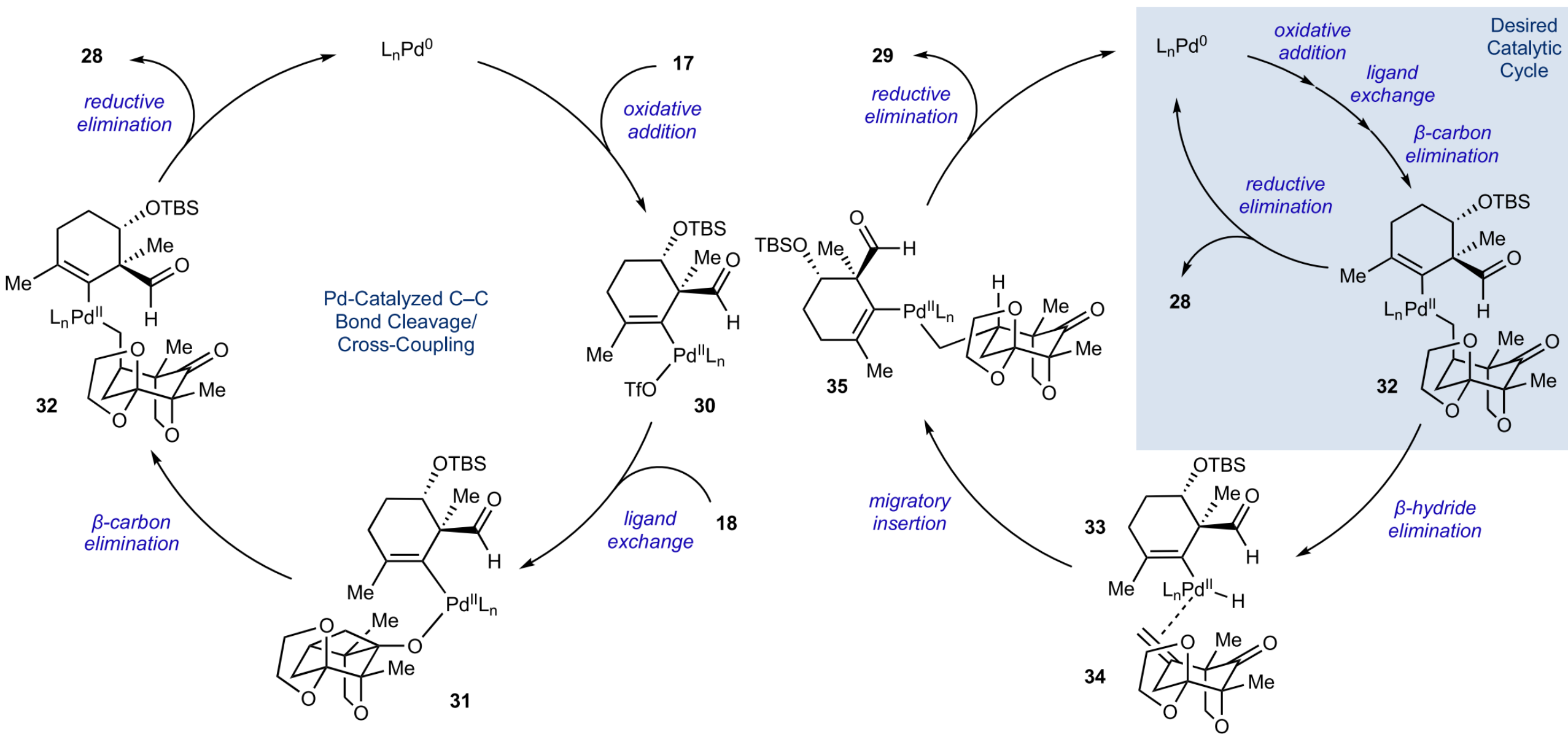


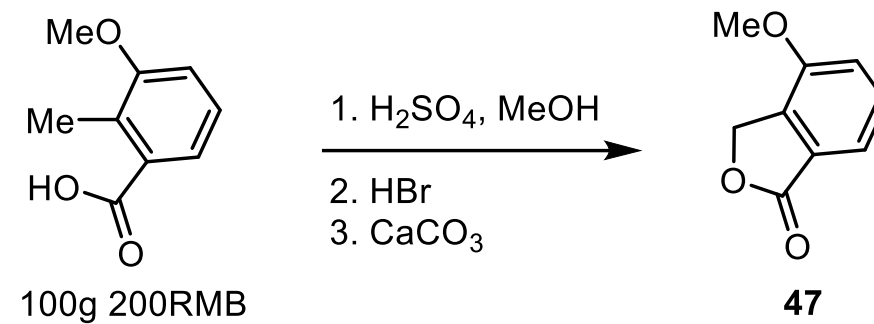
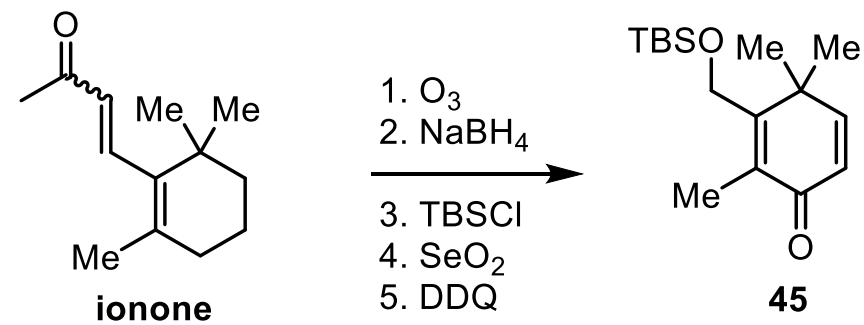
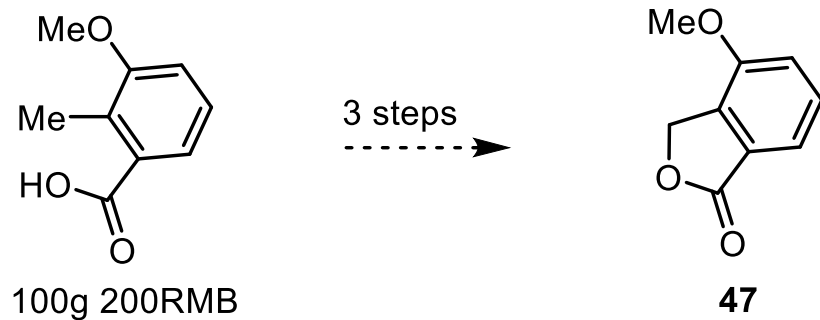
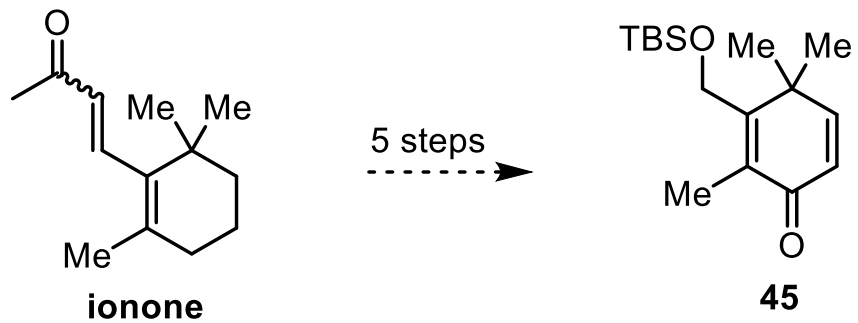
Asymmetric Synthesis

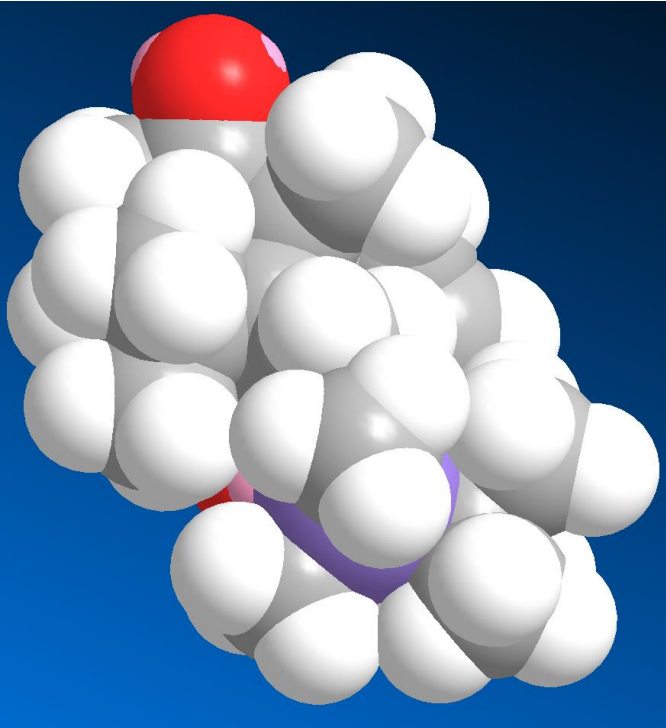


Discussion

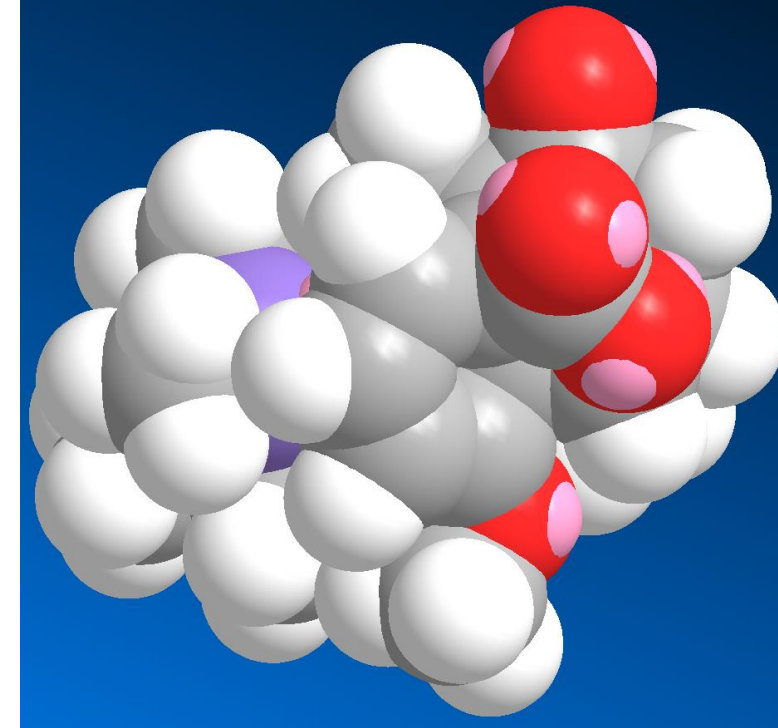
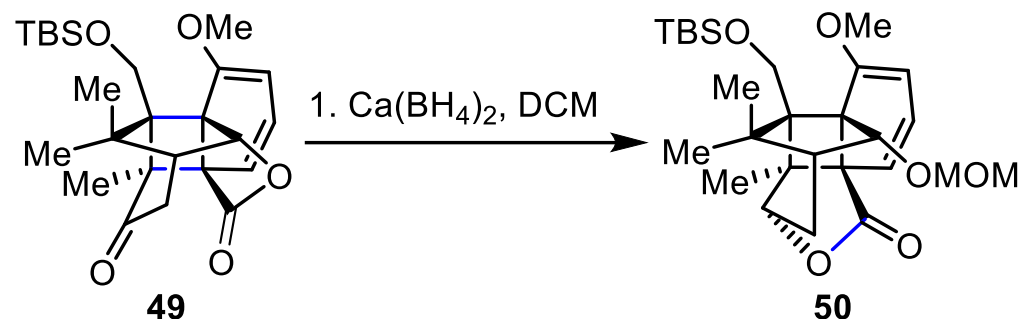








49 re-face



49 si-face

1. Atomic radius : $\text{Li}^+ < \text{Na}^+ < \text{Ca}^{2+}$
 Electric charge : $\text{Ca}^{2+} < \text{Na}^+, \text{Li}^+$
 Ability of polarize the borohydride anion
 $\text{Li}^+ > \text{Ca}^{2+} > \text{Na}^+$
2. Ability of coordinating with the carbonyl oxygen
 $\text{Li}^+ > \text{Ca}^{2+} > \text{Na}^+$
 Ability of reduction : $\text{LiBH}_4 > \text{Ca}(\text{BH}_4)_2 > \text{NaBH}_4$

