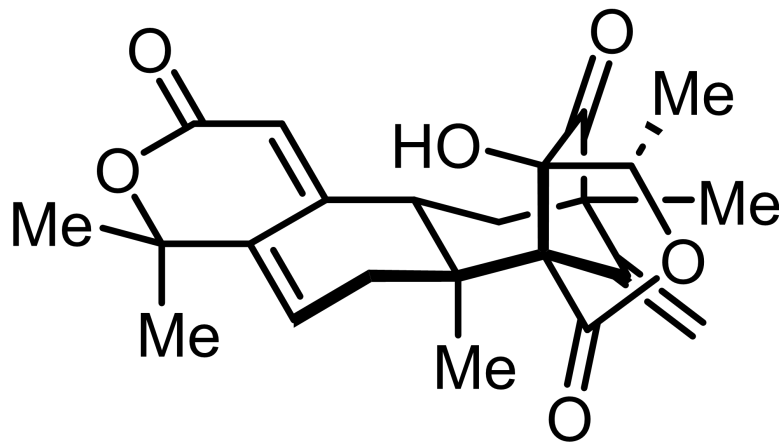


Total Synthesis of (±)-Dhilirolide U



(±)-Dhilirolide U

Wei Wang
2026.3.19

Introduction of Prof. Dr. Erick M. Carreira



- 1984** The University of Illinois at Urbana-Champaign, B.S.
Research advisor: Assistant Professor Scott E. Denmark
- 1990** Harvard University, Ph.D.
Research advisor: Professor David A. Evans
- 1990-1992** California Institute of Technology, Postdoctoral Fellow
Research advisor: Professor Peter Dervan
- Department of Chemistry, California Institute of Technology
- 1992** Assistant Professor
- 1996** Associate Professor
- 1997** Professor
- Department of Chemistry, ETH Zürich
- 1998** Professor

Introduction of (±)-Dhilirolide U

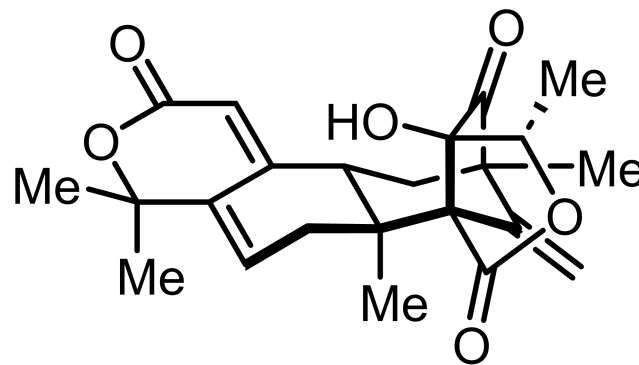
Isolation: from *Penicillium purpurogenum* AN13 by Lin and Fan et al. in 2024.

Structure:

a unique 6/6/6/5/5-fused ring system

a highly substituted bicyclo[3.2.1]octane core

an overall dense oxidation pattern



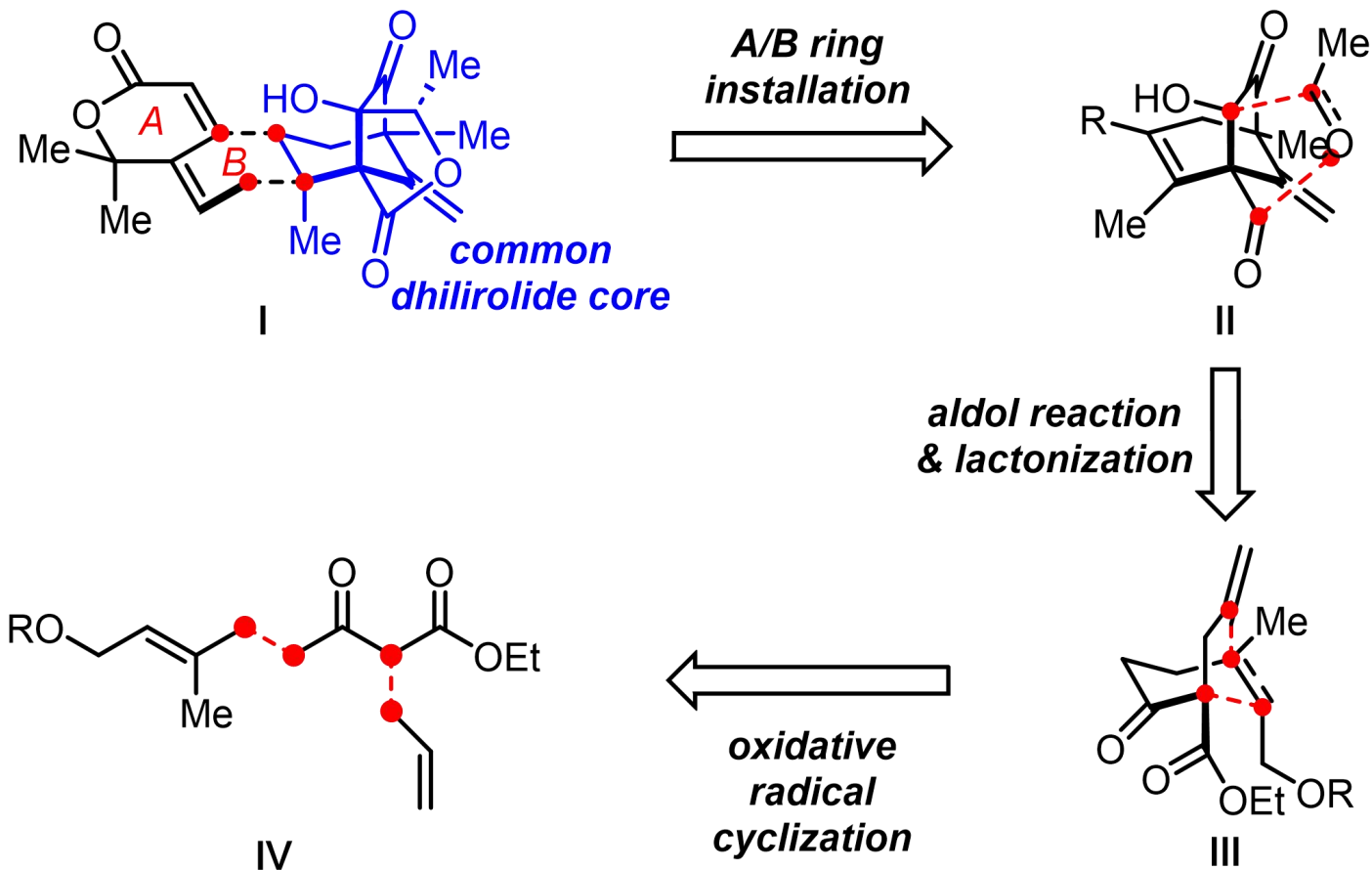
(±)-Dhilirolide U



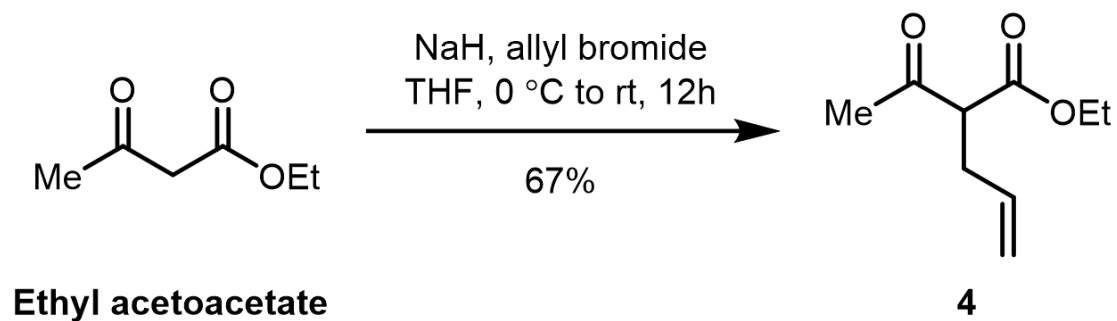
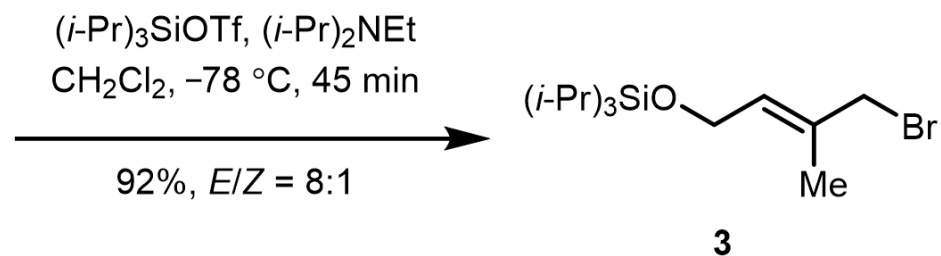
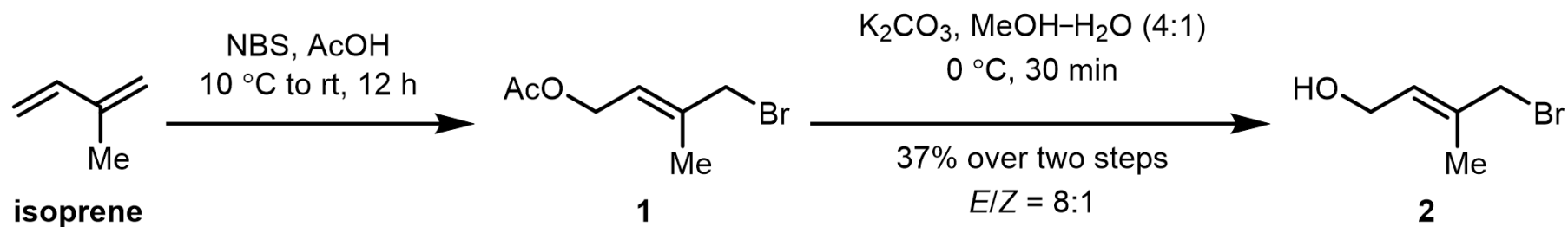
Penicillium purpurogenum fungus

Zhaolun Sun et al., *J. Am. Chem. Soc.* **2024**, 146, 30242–30251.
Henrik R. Wilke et al., *J. Am. Chem. Soc.* **2026**, 148, 6, 5916-5922.

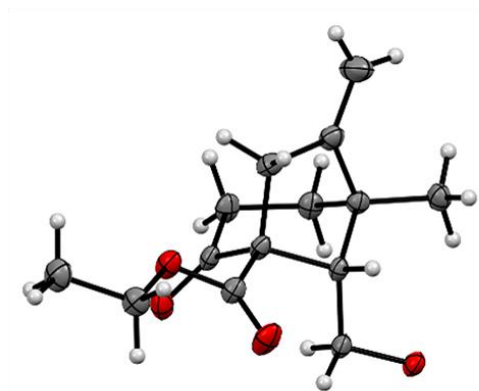
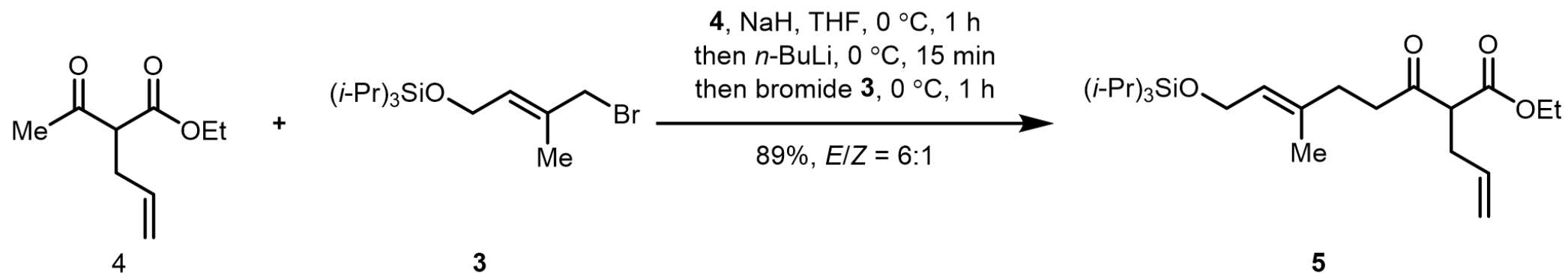
Retrosynthetic Analysis of Dhilirolide U



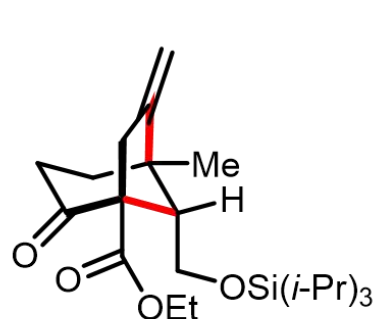
Fragment Preparation



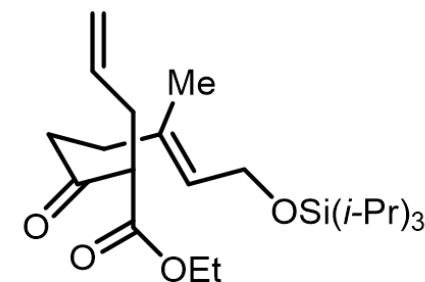
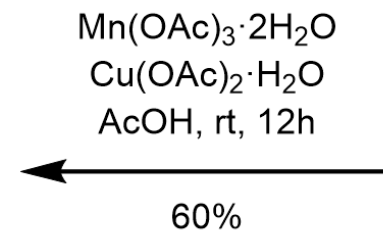
Synthesis of Bicyclo[3.2.1]octane Core



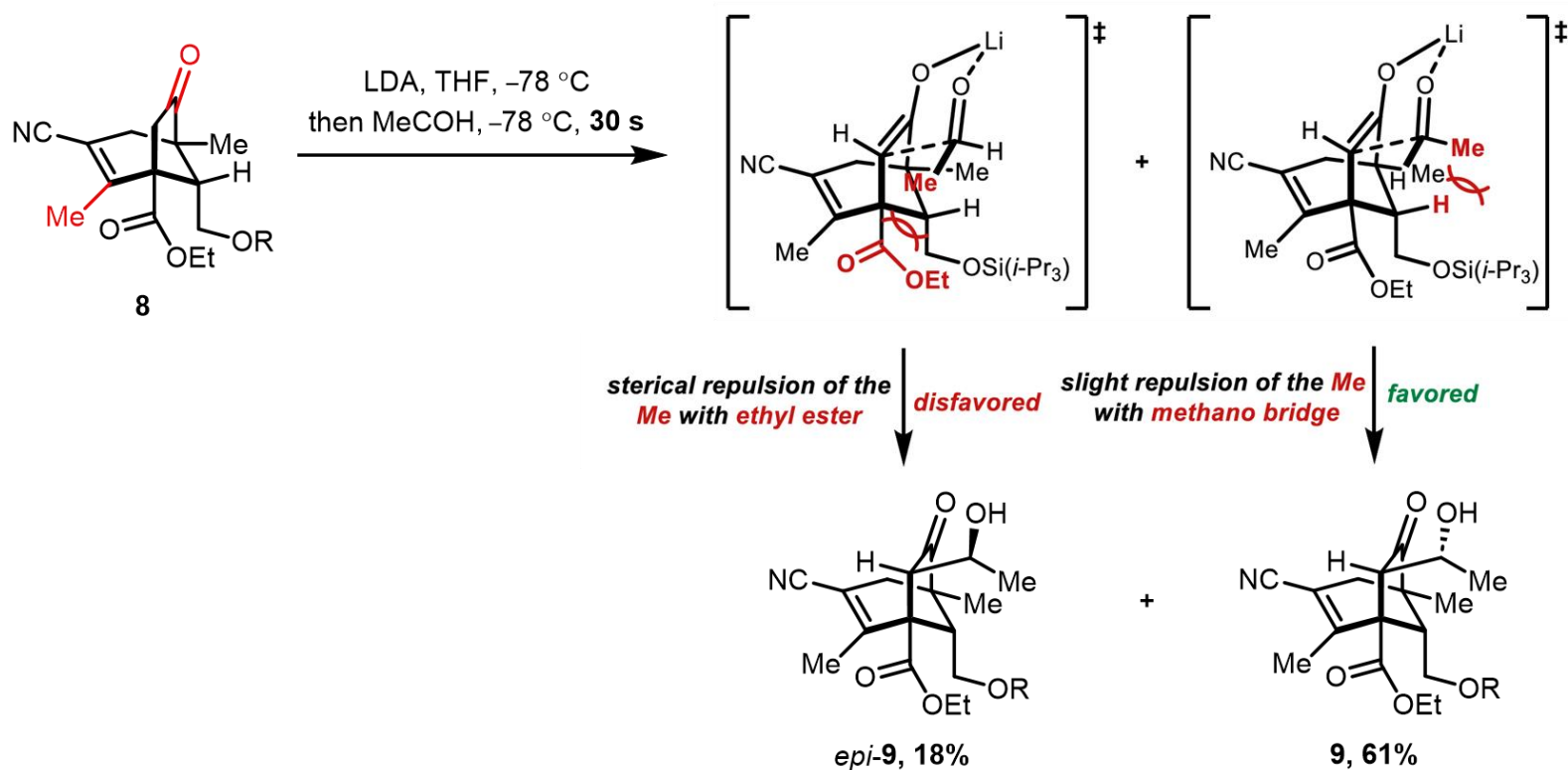
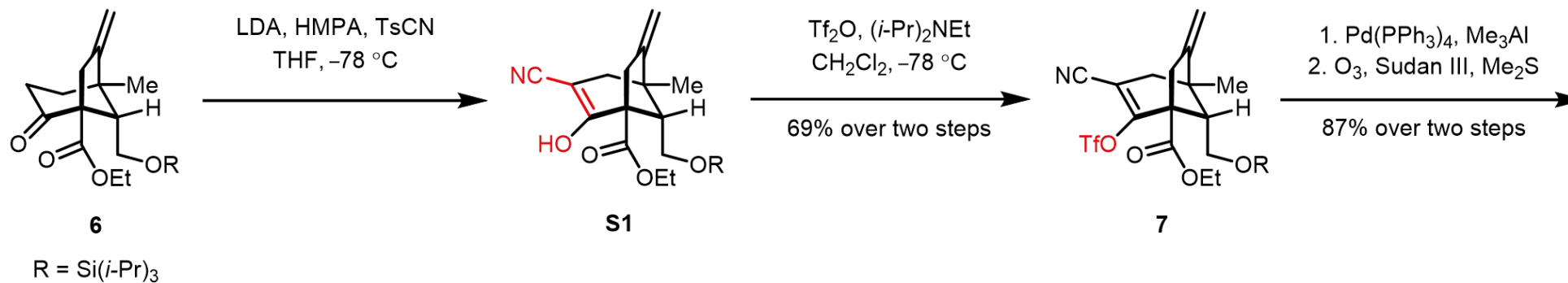
X-Ray of **6**



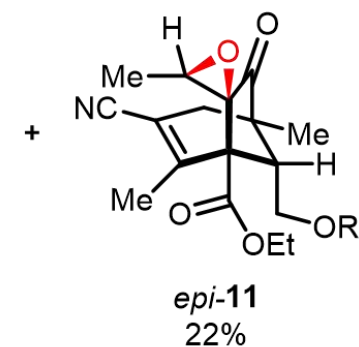
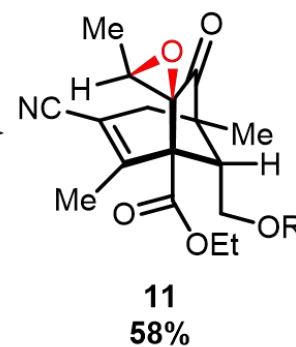
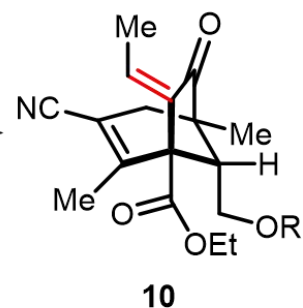
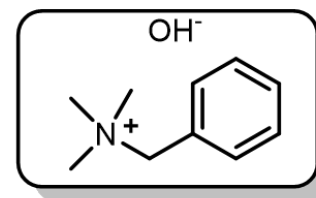
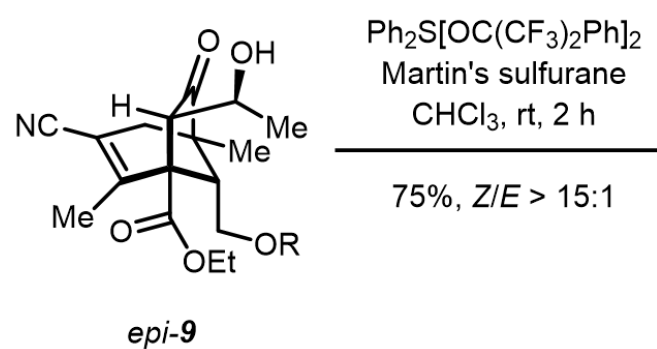
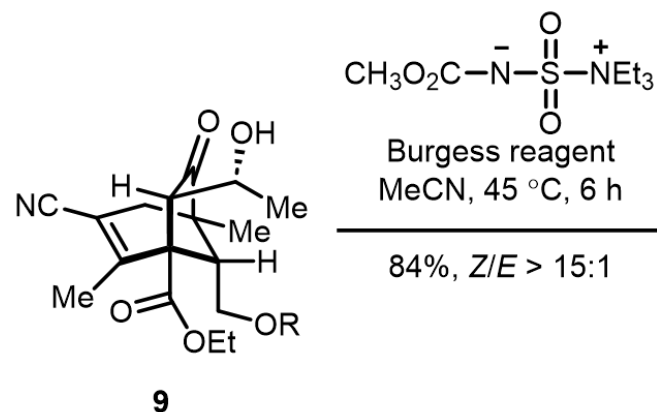
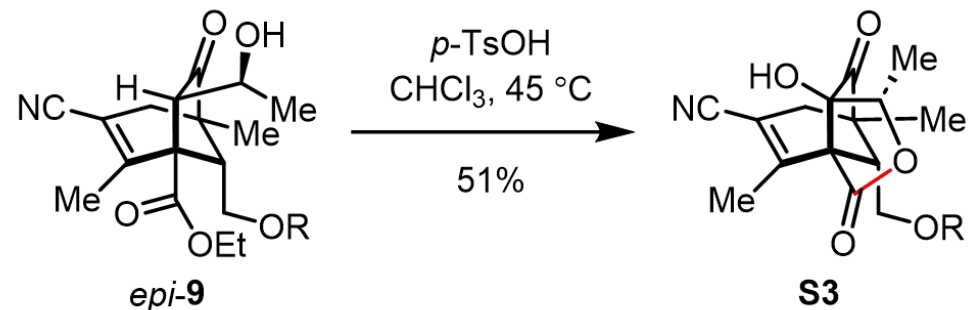
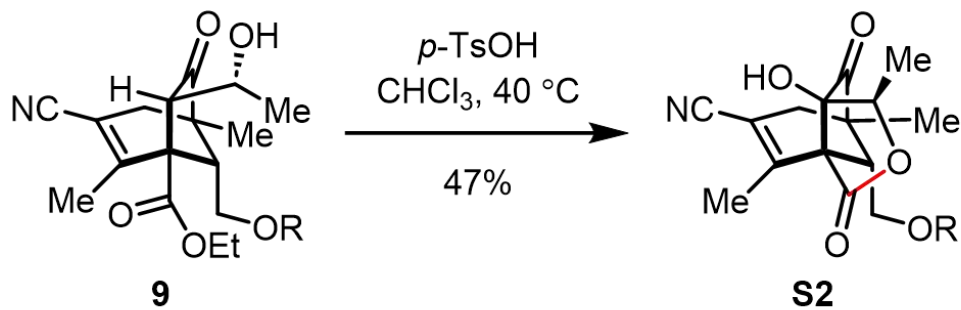
6



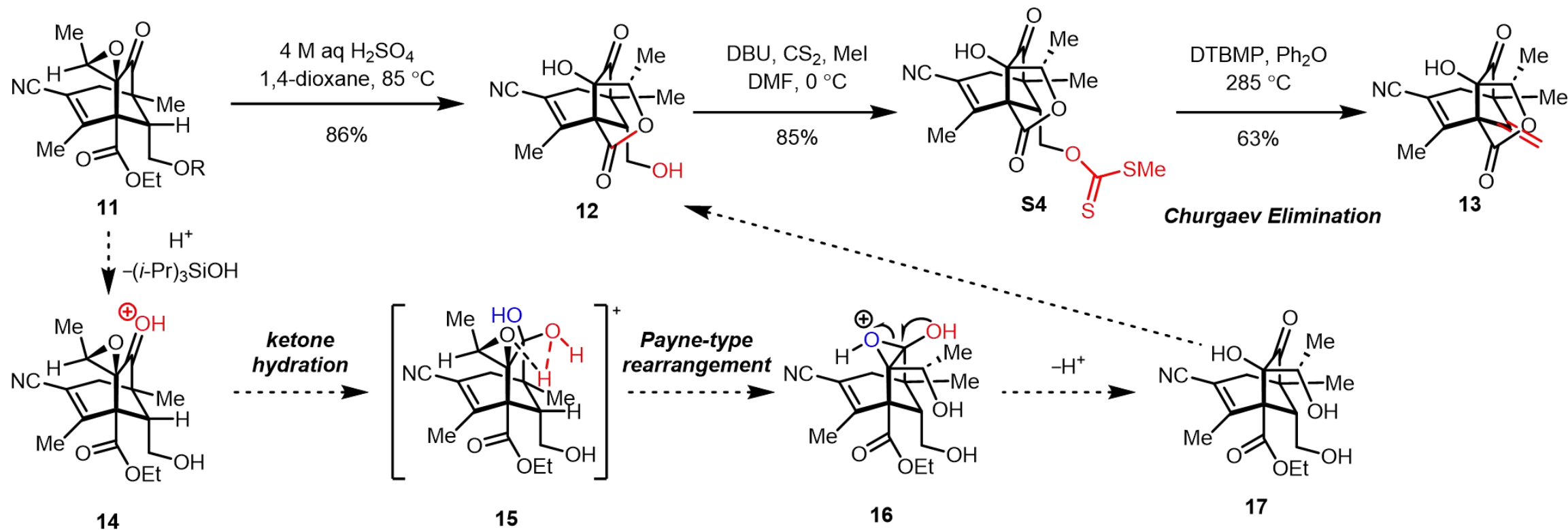
Synthesis of Lactone 13



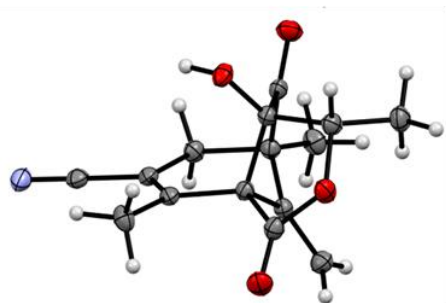
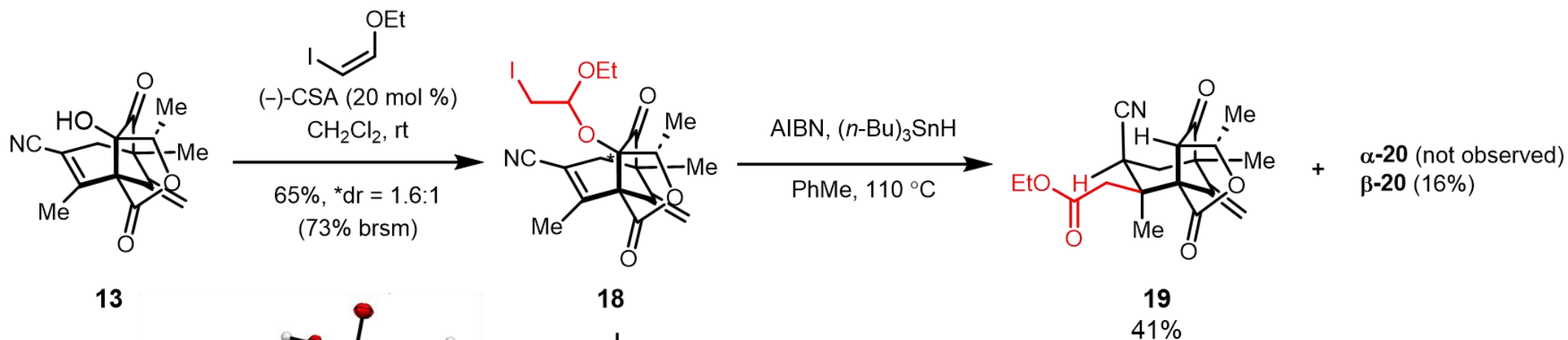
Synthesis of Lactone 13



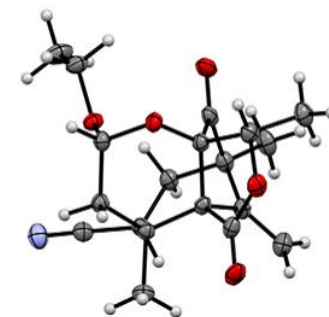
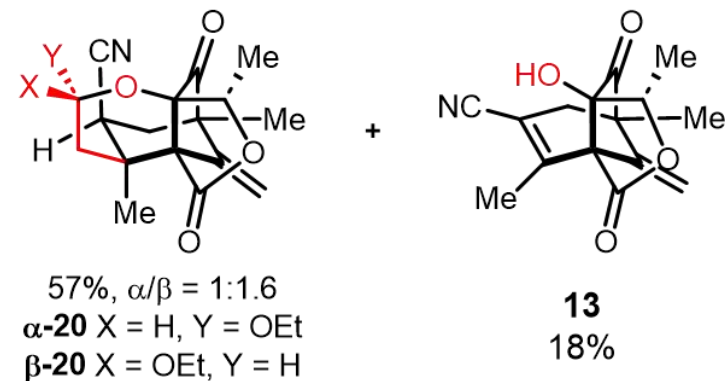
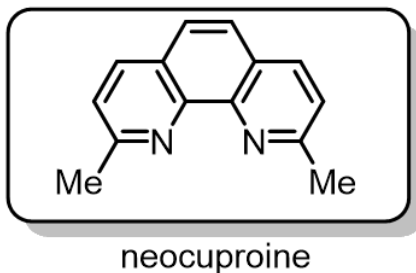
Synthesis of Lactone 13



Installation of Vicinal Quaternary Carbon Centers

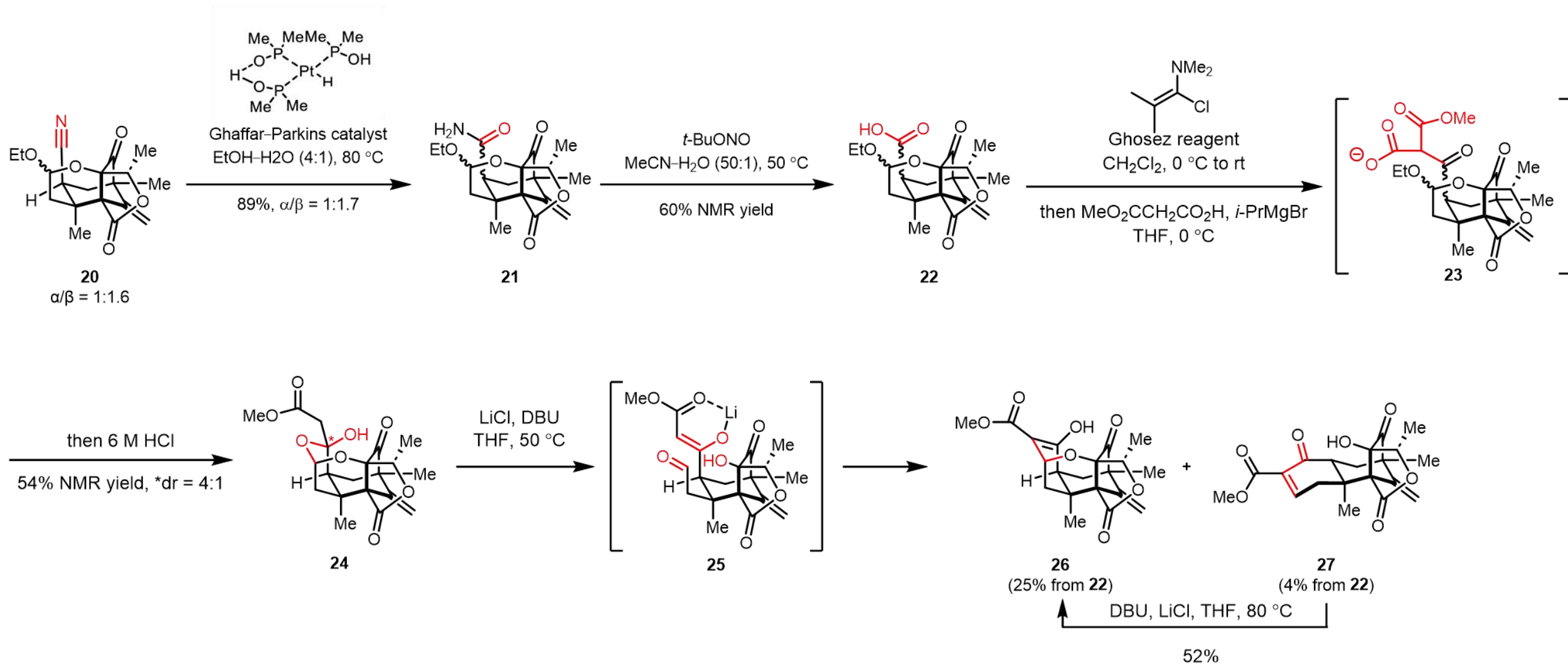


X-ray of **13**

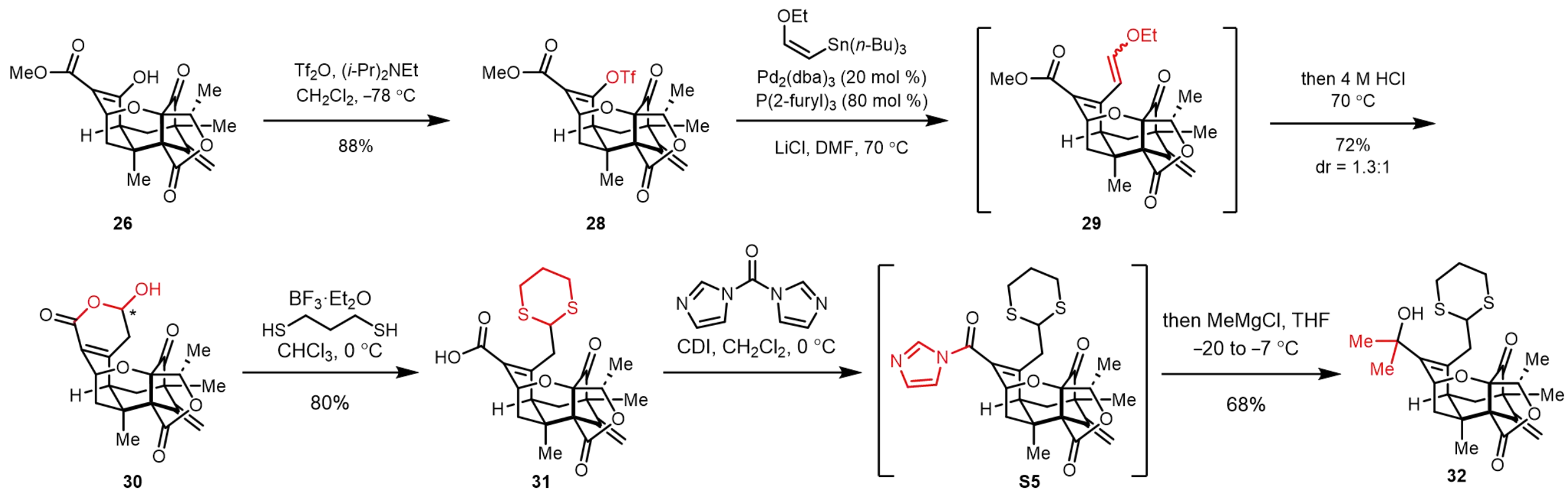


X-ray of $\alpha\text{-20}$

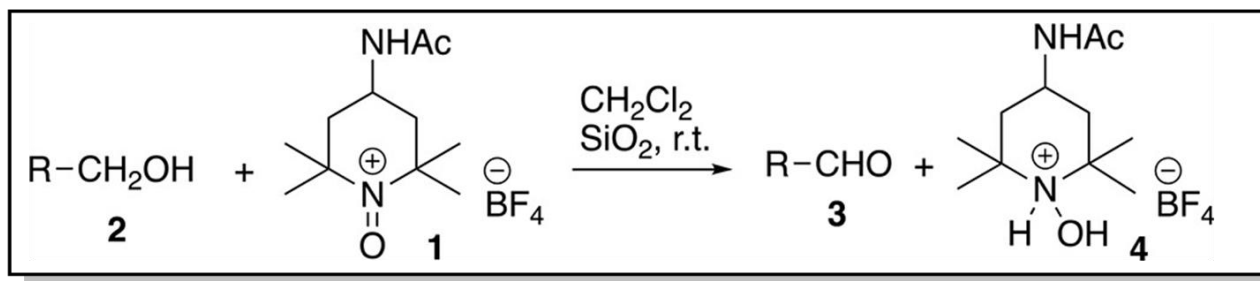
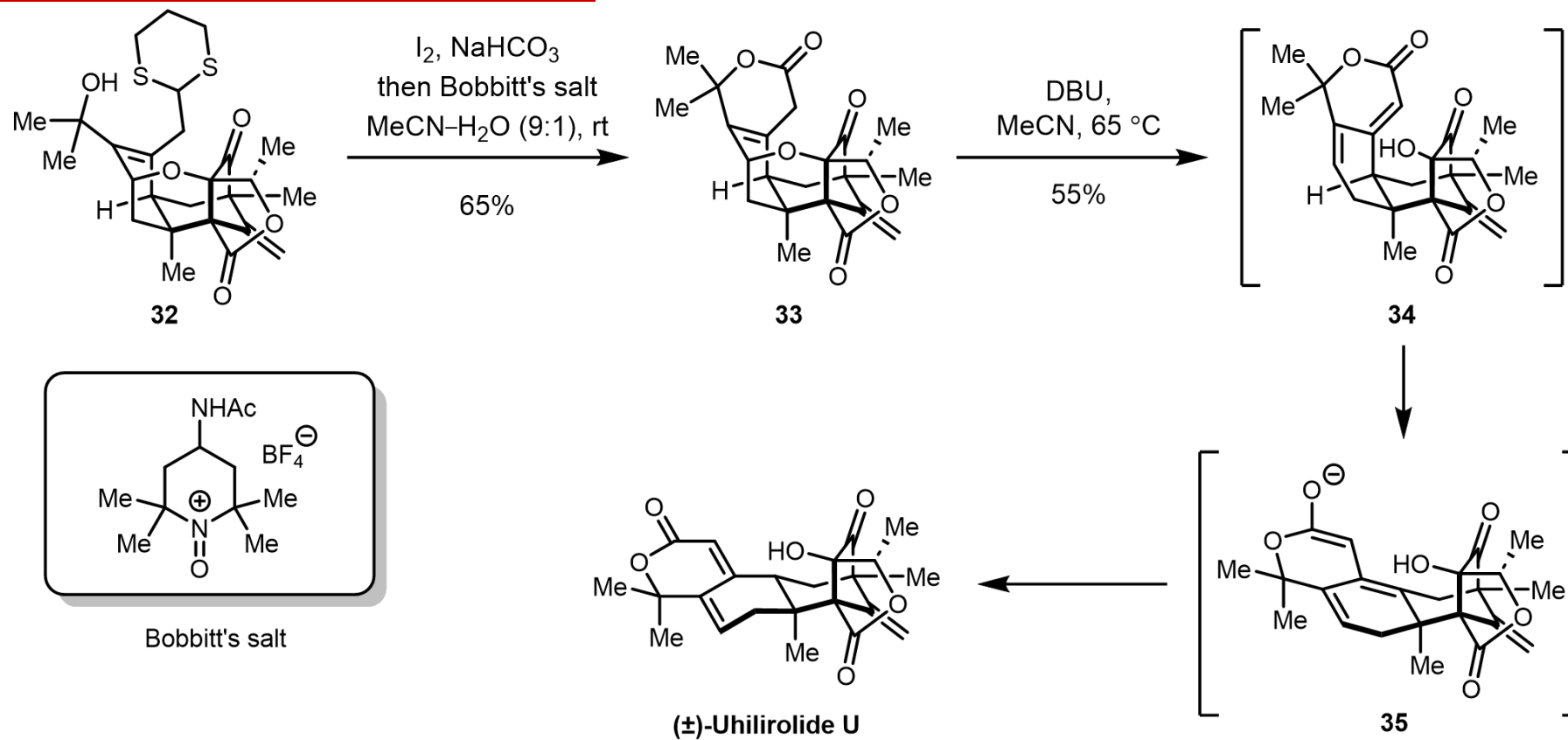
Construction of the B-Ring

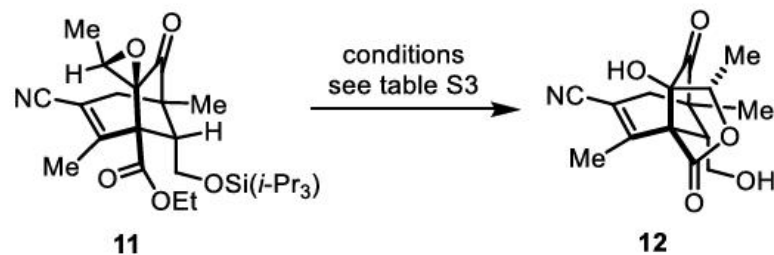


Completion of the Synthesis

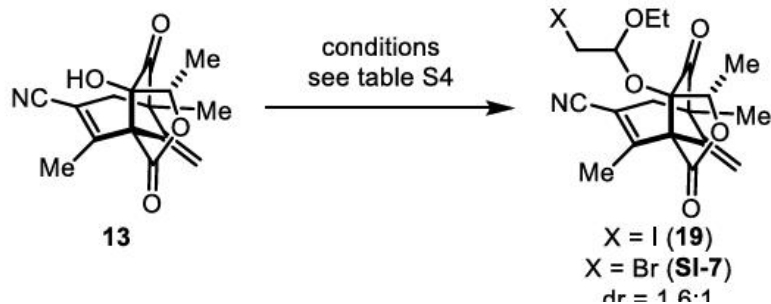


Completion of the Synthesis

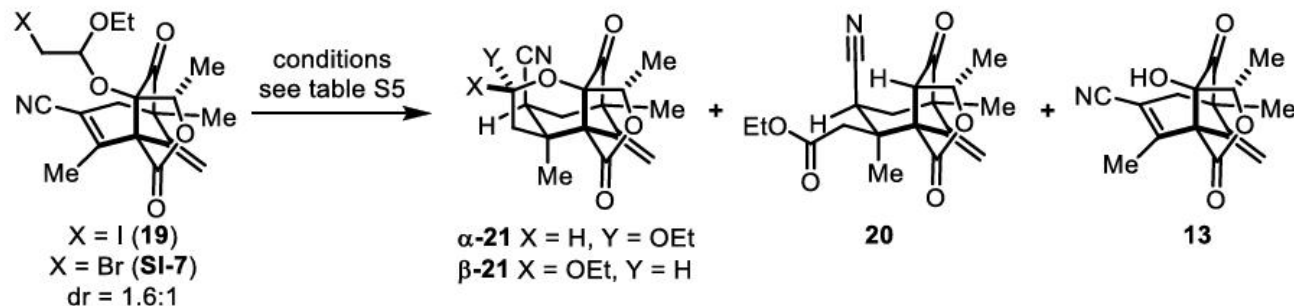




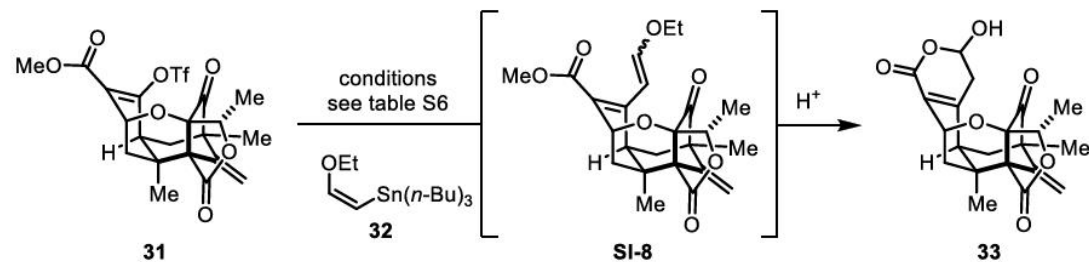
Entry	Selection of Investigated Conditions	Outcome
1	Yb(OTf) ₃ , wet THF, reflux	5%
2	HClO ₄ (30 equiv), THF–H ₂ O (5:1), 95 °C (sealed flask)	27%
3	TFA–H ₂ O–THF (1:1:2), 95 °C (sealed flask)	28%
4	TFA–H ₂ O–THF (2:1:5), 95 °C (sealed flask)	41%
5	TFA–H ₂ O–THF (1:1:2), MW, 140 °C	decomposition
6	HCl (50 equiv), THF–H ₂ O (3:1), 65 °C	33%
7	H ₂ SO ₄ (10 equiv), 1,4-dioxane–H ₂ O (3:1), 105 °C	58%
8	H ₂ SO ₄ (10 equiv), 1,4-dioxane–H ₂ O (3:1), 95 °C	70%
9	H ₂ SO ₄ (10 equiv), 1,4-dioxane–H ₂ O (3:1), 85 °C	10 mg: 83% 29.4 g-scale (isolated 86%)



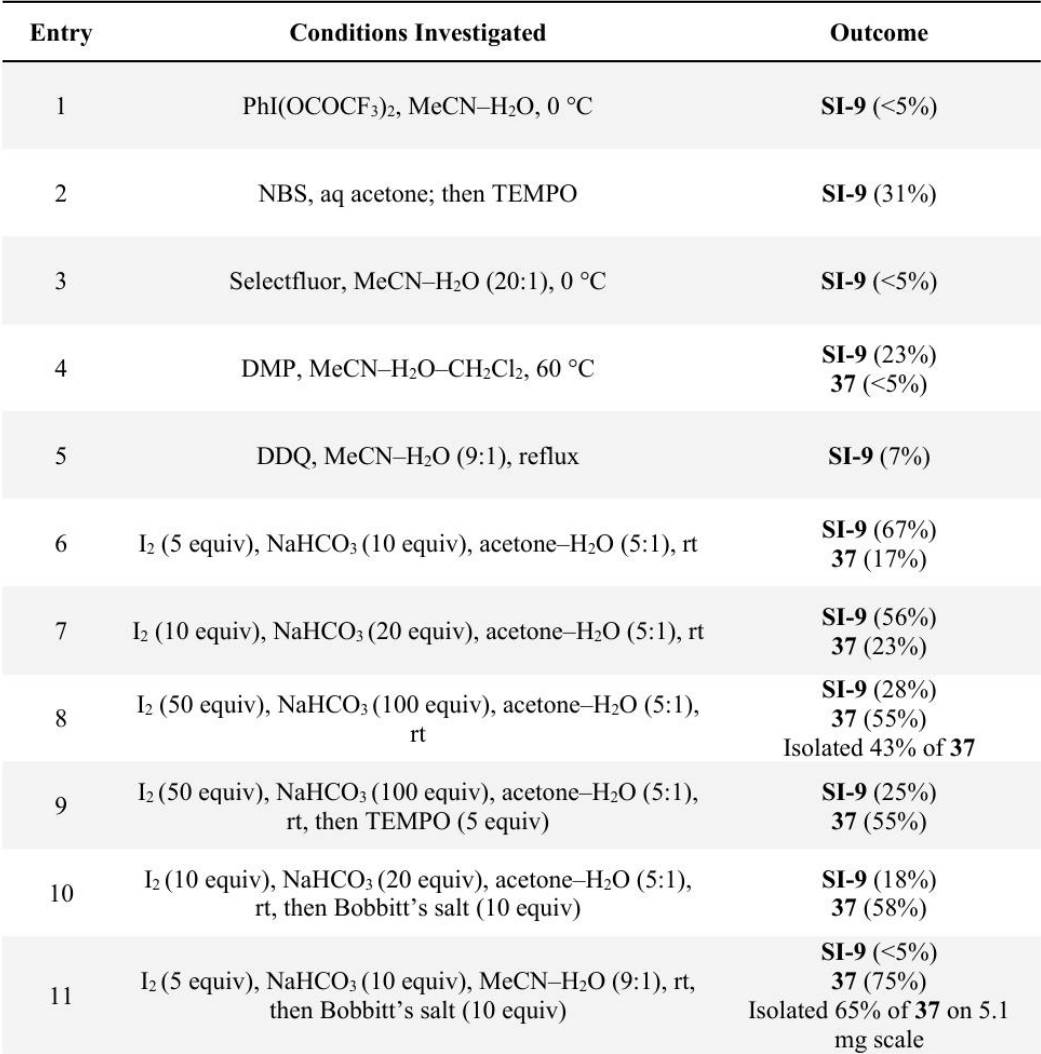
Entry	Selection of Investigated Conditions	Outcome
1	NBS (3 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), 0 °C to rt	SI-7 (24%) 13 (68%)
2	NBS (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), 0 °C to rt	SI-7 (32%) 13 (55%)
3	NBS (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), –50 to –25 °C	SI-7 (44%) 13 (38%)
4	1,2-dibromo-1-ethoxyethane, <i>N,N</i> -dimethylaniline, CH ₂ Cl ₂ (0.2 M), 0 °C to 40 °C	SI-7 (17%) 13 (70%)
5	1,3-Dibromo-5,5-dimethylhydantoin (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), –20 °C to rt	SI-7 (7%) 13 (79%)
6	Dibromoisocyanuric acid (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), –20 °C to rt	SI-7 (<5%)
7	<i>N</i> -Bromosaccharin (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), –20 °C to rt	SI-7 (<5%)
8	NIS (5 equiv), ethyl vinyl ether (10 equiv), CH ₂ Cl ₂ (0.2 M), 0 °C to rt	messy reaction
9	(–)-CSA (20 mol %), (<i>Z</i>)-1-bromo-2-ethoxyethene (3 equiv), CH ₂ Cl ₂ (0.2 M), rt	SI-7 (52%) 13 (43%)
10	(–)-CSA (20 mol %), (<i>Z</i>)-1-bromo-2-ethoxyethene (2.0 equiv), CH ₂ Cl ₂ (0.4 M), rt	SI-7 (85%) 13 (4%) on 9.27 mmol scale
11	(–)-CSA (20 mol %), (<i>Z</i>)-1-iodo-2-ethoxyethene (2.0 equiv), CH ₂ Cl ₂ (0.4 M), rt	19 (68%, isolated 65%) 13 (13%, isolated 11%) on 403 μmol scale



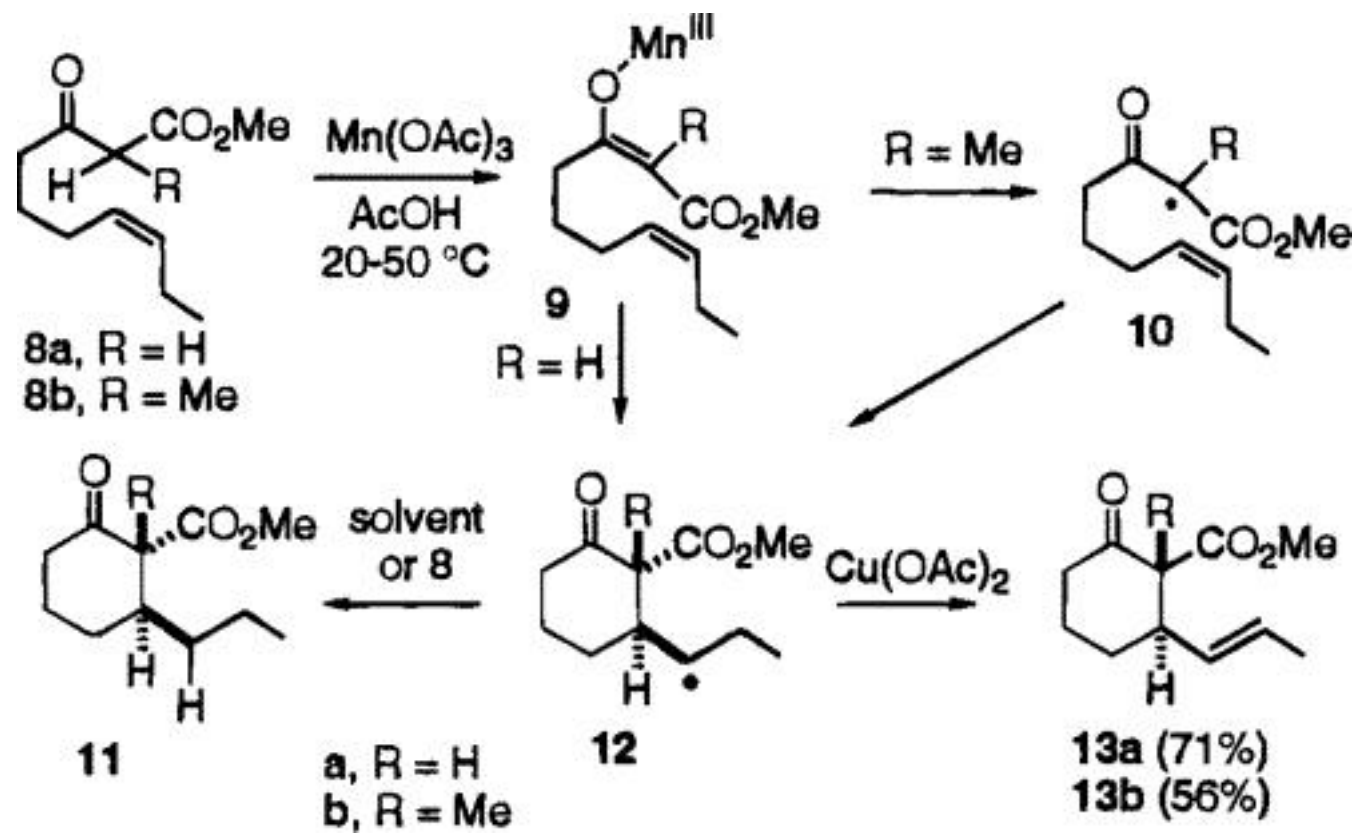
Entry	Selection of Investigated Conditions	Outcome
1	19 ($\text{X} = \text{I}$), AIBN (50 mol%), (<i>n</i> -Bu) ₃ SnH (5.0 equiv), PhMe, 110 °C, sealed flask	$\alpha\text{-21}$ (not observed) $\beta\text{-21}$ (19%, isolated 16%) 20 (45%, isolated 41%) (messy reaction)
2	19 ($\text{X} = \text{I}$), AIBN (50 mol%), Ph ₃ SnH (5.0 equiv), PhMe, 110 °C, sealed flask	$\alpha\text{-21}$ (not observed) $\beta\text{-21}$ (11%) 20 (29%) (messy reaction)
3	19 ($\text{X} = \text{I}$), Et ₃ B (50 mol %), (<i>n</i> -Bu) ₃ SnH (5.0 equiv), PhMe, -78 °C	$\alpha\text{-21}$ (7%) $\beta\text{-21}$ (17%) (messy reaction)
4	19 ($\text{X} = \text{I}$), Ir[dF(CF ₃)ppy] ₂ (dtbbpy)PF ₆ (10 mol %), (<i>i</i> -Pr) ₂ NEt, H ₂ O (20 equiv), MeCN, rt, blue LED	$\alpha\text{-21}$ (18%) $\beta\text{-21}$ (34%) 13 (4%)
5	19 ($\text{X} = \text{I}$), NiCl ₂ ·glyme (20 mol %), neocuproine (24 mol %), zinc dust, MeOH, reflux	$\alpha\text{-21}$ (<5%) $\beta\text{-21}$ (<5%) 13 (11%)
6	19 ($\text{X} = \text{I}$), NiCl ₂ ·glyme (20 mol %), neocuproine (24 mol %), zinc nanopowder (Sigma Aldrich), MeOH, 50 °C	$\alpha\text{-21}$ (25%, isolated 22%) $\beta\text{-21}$ (41%, isolated 37%) 13 (15%)
7	19 ($\text{X} = \text{I}$), NiCl ₂ ·glyme (20 mol %), neocuproine (24 mol %), Rieke zinc in THF, MeOH, 50 °C to 60 °C	$\alpha\text{-21}$ (7%) $\beta\text{-21}$ (12%) 13 (65%)
8	19 ($\text{X} = \text{I}$), NiCl ₂ ·glyme (20 mol %), neocuproine (24 mol %), Rieke zinc, MeOH, 50 °C	$\alpha\text{-21}$ (24%, isolated 22%) $\beta\text{-21}$ (38%, isolated 35%) 13 (21%, isolated 18%) on 5.28 mmol scale
9	SI-7 ($\text{X} = \text{Br}$), NiCl ₂ ·glyme (20 mol %), neocuproine (24 mol %), Rieke zinc, MeOH, reflux	$\alpha\text{-21}$ (16%) $\beta\text{-21}$ (23%) 13 (42%)



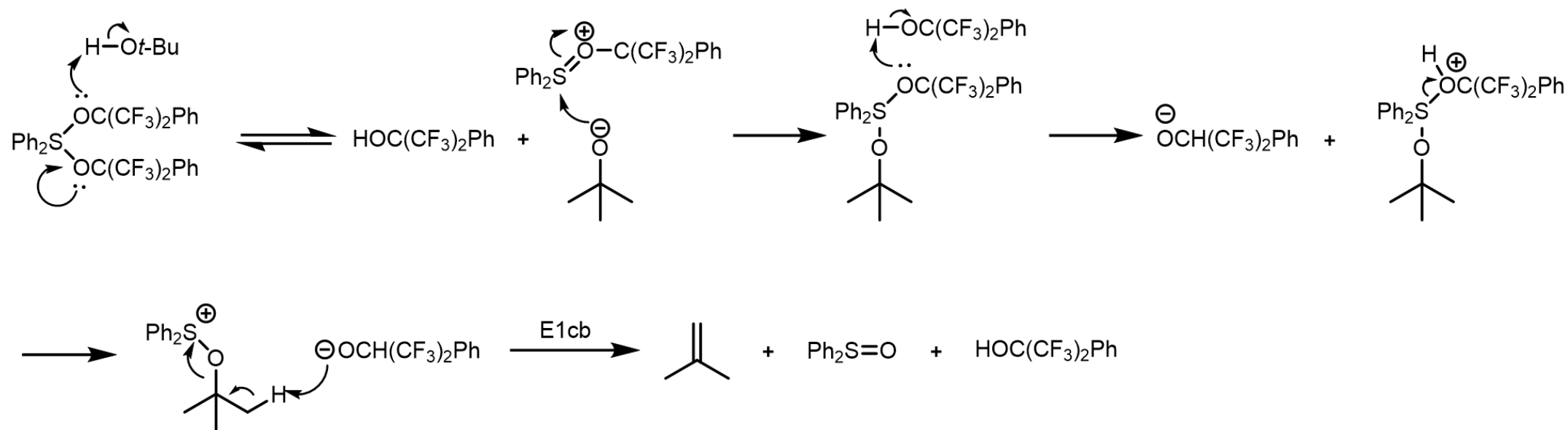
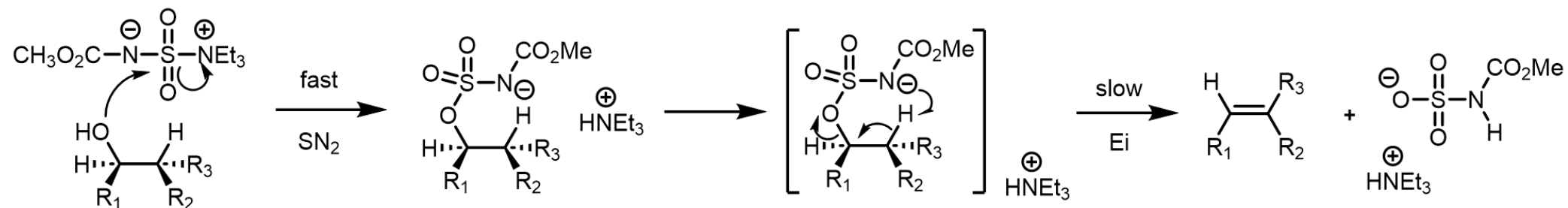
Entry	Conditions Investigated	Outcome
1	$Pd(PPh_3)_4$ (20 mol %), LiCl (5 equiv), 32 (5 equiv), THF, 65 °C	SI-8 (<10%)
2	$Pd_2(dba)_3$ (20 mol %), $AsPh_3$ (80 mol %), LiCl (5 equiv), 32 (5 equiv), THF, 65 °C	SI-8 (<10%)
3	$Pd_2(dba)_3$ (20 mol %), $AsPh_3$ (80 mol %), LiCl (5 equiv), 32 (5 equiv), 1,4-dioxane, 90 °C	SI-8 (28%)
4	$Pd(OAc)_2$ (20 mol %), XPhos (40 mol %), CsCl (10 equiv), Cs_2CO_3 (50 mol %), 32 (5 equiv), 1,4-dioxane, 90 °C	SI-8 (55%) Isolated 38%
5	$Pd(OAc)_2$ (20 mol %), XPhos (40 mol %), CsCl (10 equiv), Cs_2CO_3 (50 mol %), 32 (5 equiv), 1,4-dioxane, 90 °C; then 4 M aq HCl, 60 °C	33 (47%) Isolated 40%
6	$Pd_2(dba)_3$ (20 mol %), $P(2-furyl)_3$ (80 mol %), 32 (5 equiv), 1,4-dioxane, 75 °C	SI-8 (67%)
7	$Pd_2(dba)_3$ (20 mol %), $P(2-furyl)_3$ (80 mol %), 32 (5 equiv), LiCl (5 equiv), 1,4-dioxane, 75 °C	SI-8 (76%)
8	$Pd_2(dba)_3$ (20 mol %), $P(2-furyl)_3$ (80 mol %), 32 (5 equiv), DMF, 75 °C	SI-8 (81%) Isolated 55%
9	$Pd_2(dba)_3$ (20 mol %), $P(2-furyl)_3$ (80 mol %), 32 (5 equiv), 1,4-dioxane, 80 °C; then 4 M aq HCl, 60 °C	33 (72%) Isolated 60%
10	$Pd_2(dba)_3$ (20 mol %), $P(2-furyl)_3$ (80 mol %), 32 (5 equiv), LiCl (5 equiv), DMF, 70 °C; then 4 M aq HCl, 70 °C	33 (76%) Isolated 72% (from 52.3 mg of 33)



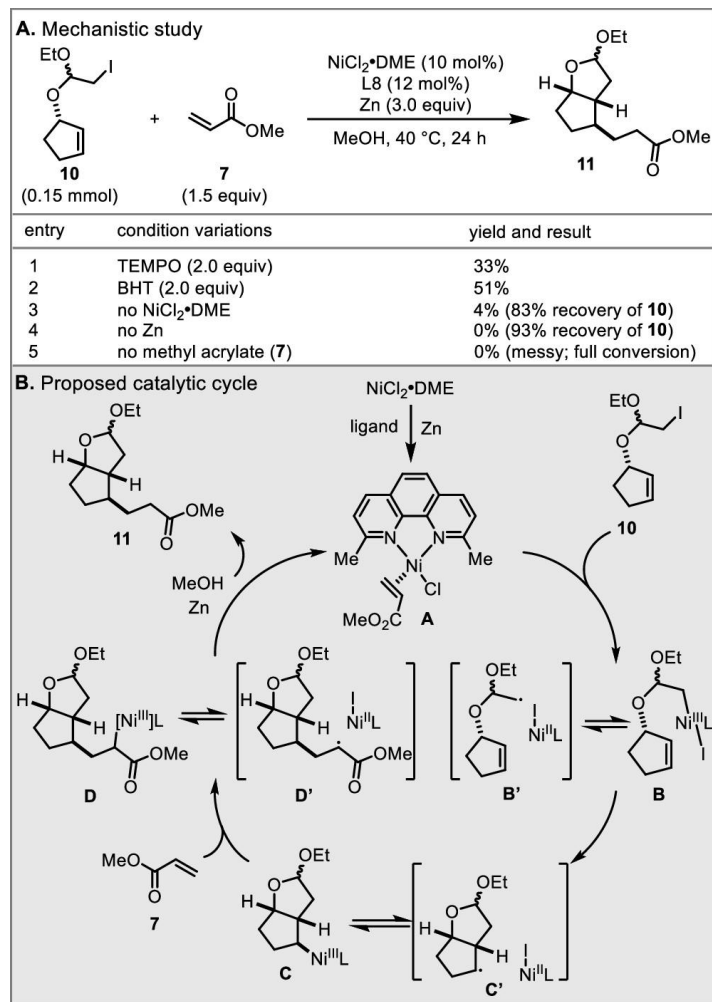
Manganese(III)-Based Oxidative Free-Radical Cyclizations



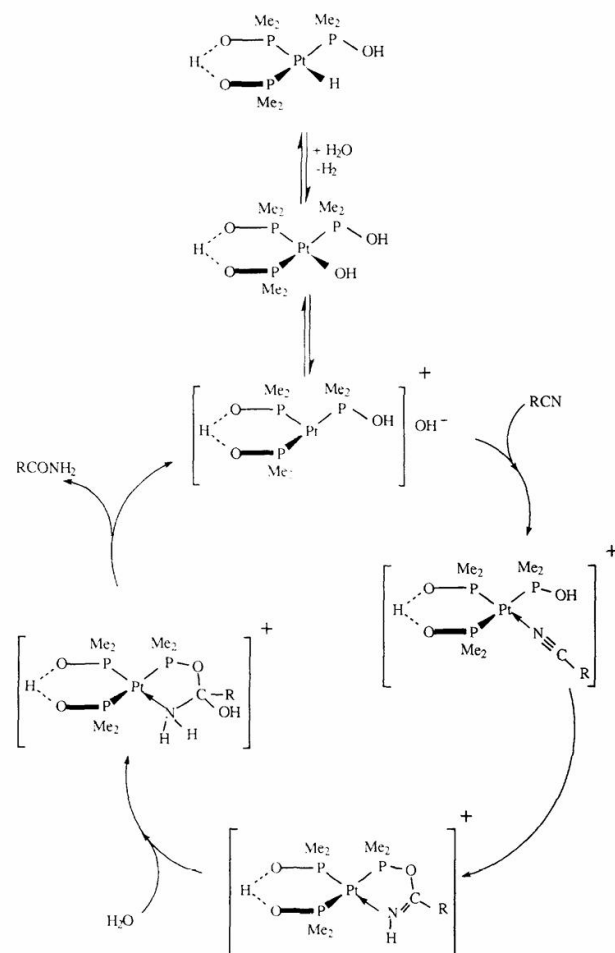
Burgess reagent and Martin's sulfurane



Ueno–Stork cyclization

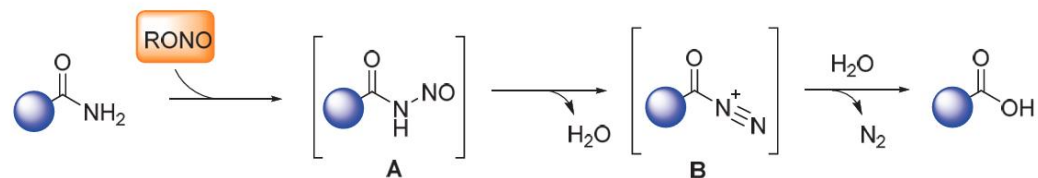
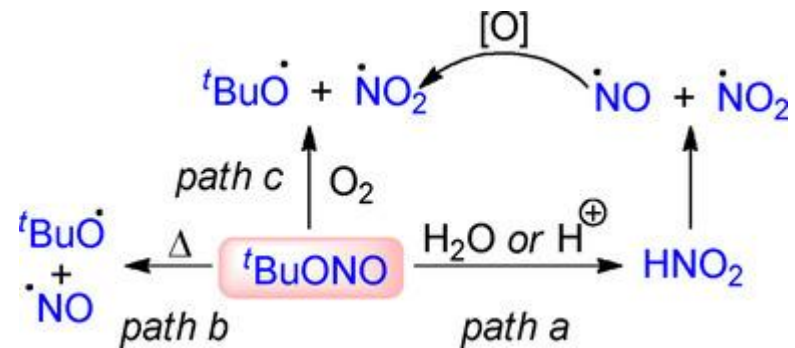


Ghaffar-Parkins catalyst

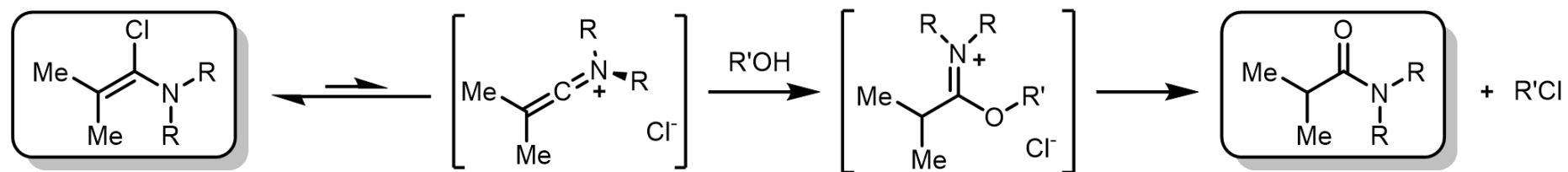


Talit Ghaffar and Adrian W. Parkins, *Tetrahedron Lett.* **1995**, 36, 8657–8660.

Amide Hydrolysis Reaction Using tert-Butyl Nitrite



The Replacement of a Hydroxyl Group by Halogen



R' = Alkyl or carboxylic acids